



Reversible Bonds-Manipulation to Access Functionalized Molecules

Tristan Delcaillau

Zhu group

LSPN - EPFL

10/11/2022

Prof. Bill Morandi

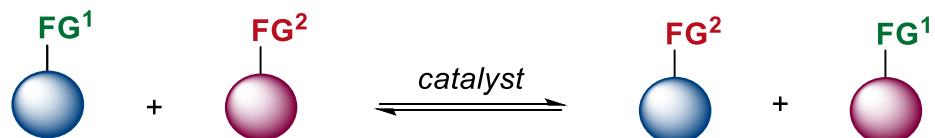
ETH zürich



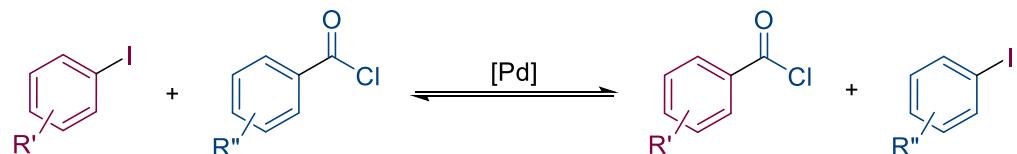
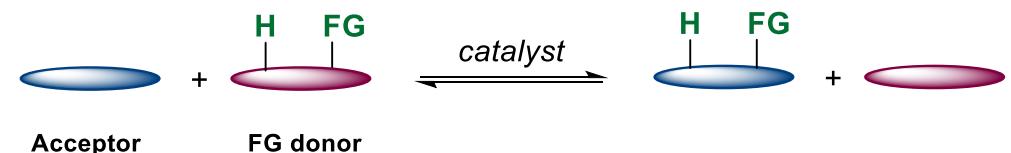
- Born in Fribourg, Switzerland
- PhD in Carreira group, ETH – Switzerland (2008-2012)
- Postdoc in Grubbs group, Caltech – US (2012-2014)
- Independent leader at MPI – Germany (2014-2018)
- Prof. at ETH Zürich – Switzerland (2018-)

Chemistry of Morandi group

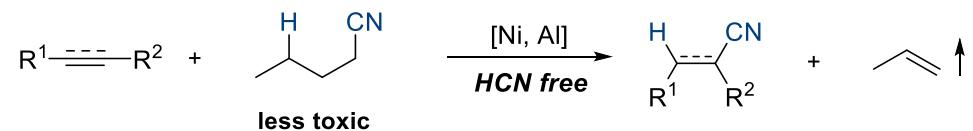
Functional group metathesis



Shuttle Catalysis

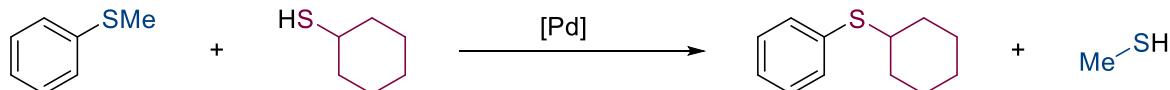


Nat. Chem. 2018, 10, 1016

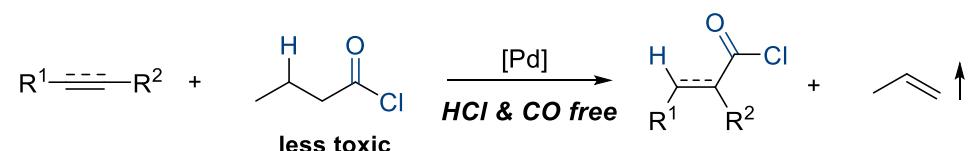


Science 2016, 351, 832–836

Single-bond metathesis



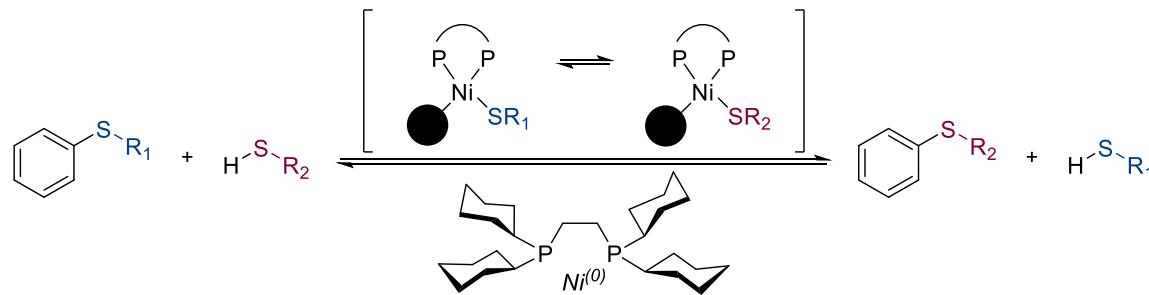
Science, 2017, 356, 1059–1063



Nat. Chem. 2017, 9, 1105–1109

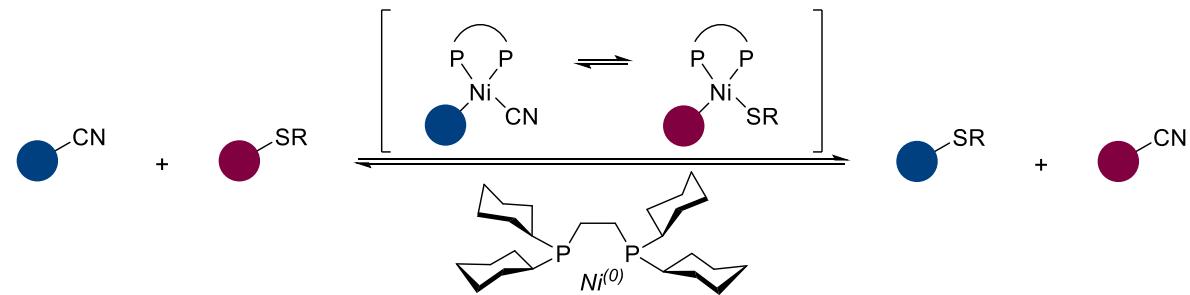
Overview

Aryl thioether metathesis



Angew. Chem. Int. Ed. **2020**, *59*, 2110–2114.

Functional group metathesis between ArCN and ArSR



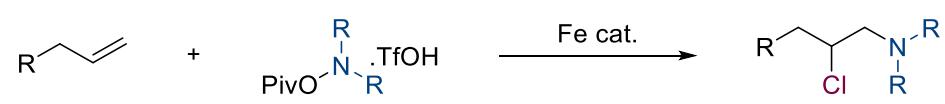
J. Am. Chem. Soc. **2021**, *143*, 3723–3728.
Chem. Eur. J. **2021**, *27*, 11823–11826.
Org. Lett. **2021**, *23*, 18, 7018–7022.

C(sp²)-S / C(sp³)-S metathesis



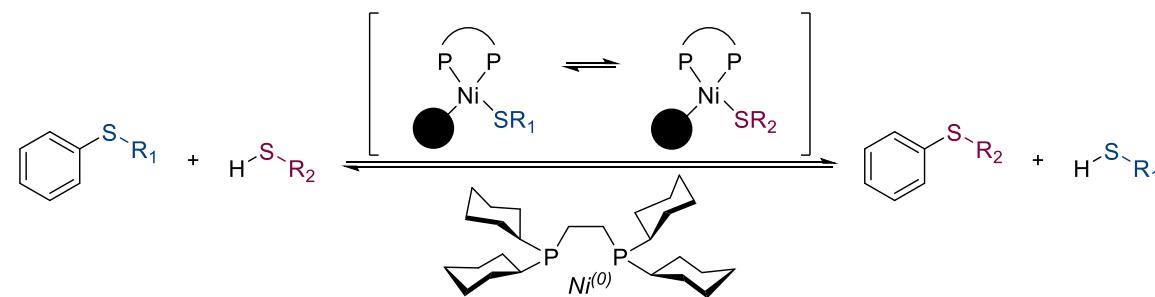
ACS Catal. **2022**, *12*, 6081–6091.

Amination of alkenes



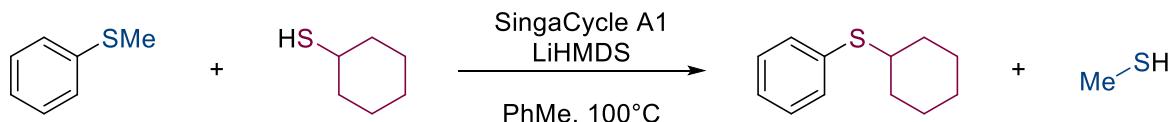
Angew. Chem. Int. Ed. **2020**, *59*, 21064.
Science **2018**, *362*, 434–439.

Reversible Inter- and Intramolecular Aryl Thioether Metathesis

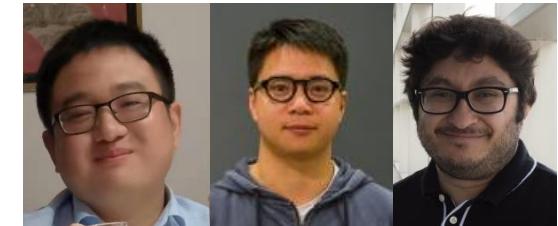
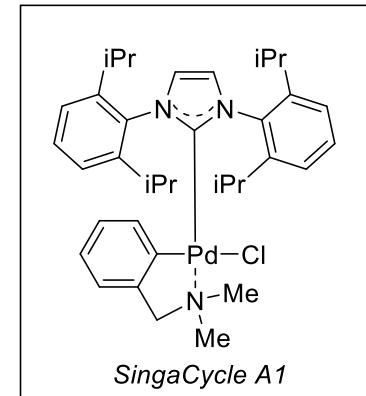


Reversible Inter- and Intramolecular Aryl Thioether Metathesis

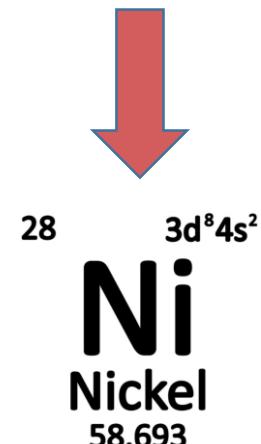
Aryl thioether metathesis



Science, 2017, 356, 1059–1063

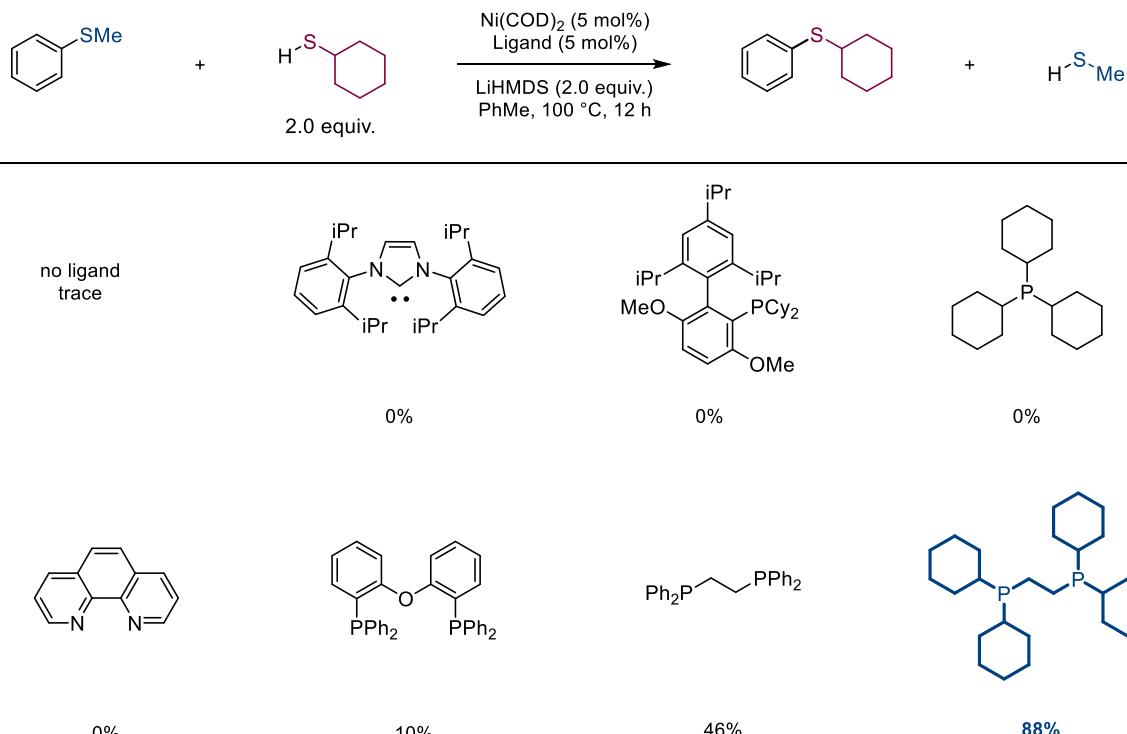


- Limited scope
- Expensive Pd cat.



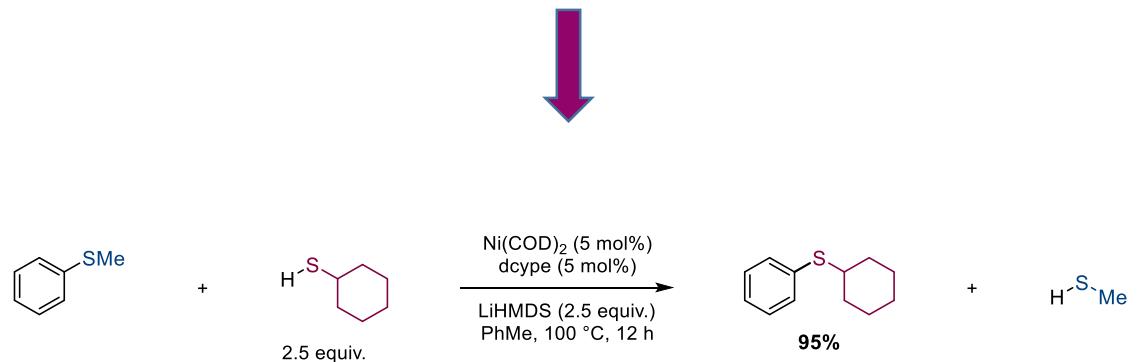
Reversible Inter- and Intramolecular Aryl Thioether Metathesis

Optimization

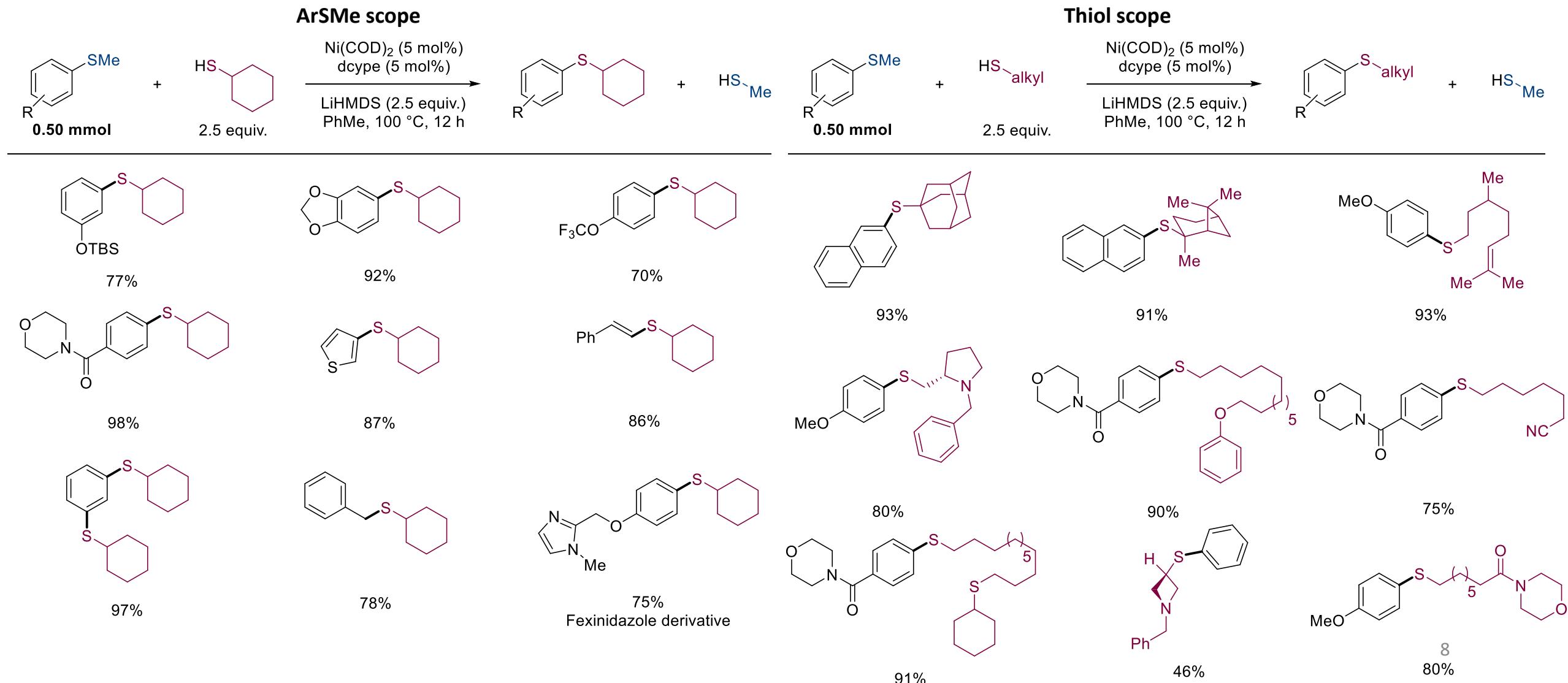


Further optimization

- Nickel loading
- Solvent
- Ligand loading
- Base
- Temperature
- Amount of thiol
- Time
- Amount of base

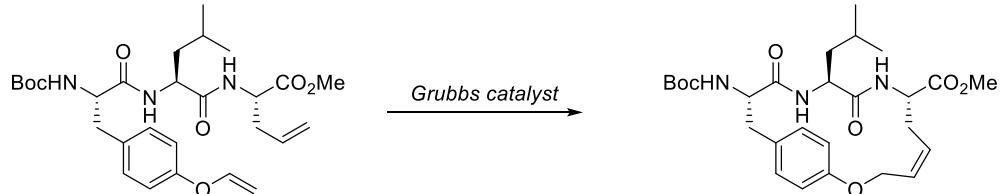


Reversible Inter- and Intramolecular Aryl Thioether Metathesis

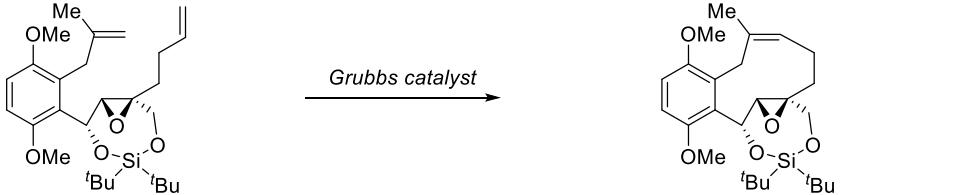


Reversible Inter- and Intramolecular Aryl Thioether Metathesis

Olefin metathesis



Chem. Sci. 2015, 6, 4561–4569



Intermediate en route to (+)-Clavilactone A
Org. Lett. 2013, 15, 5582–5585

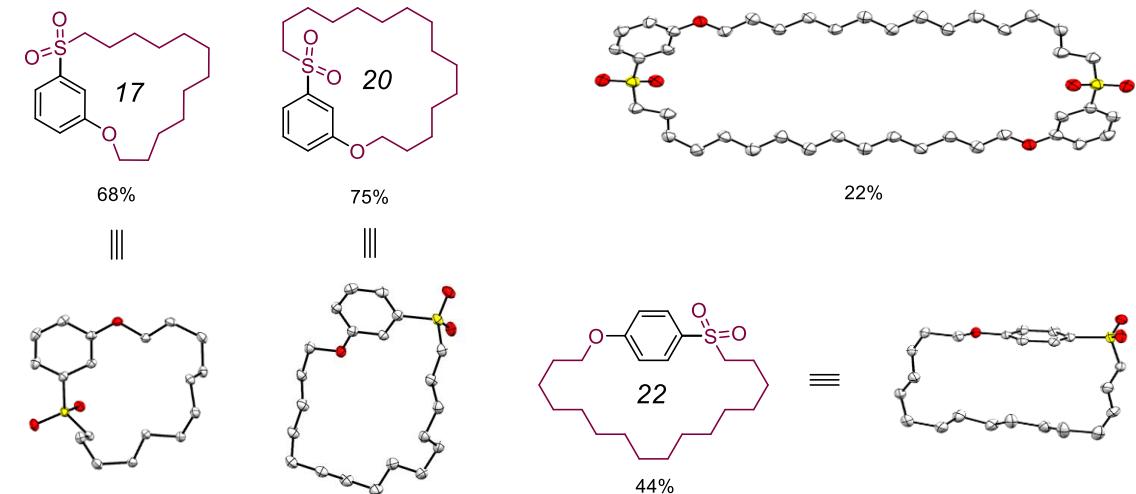
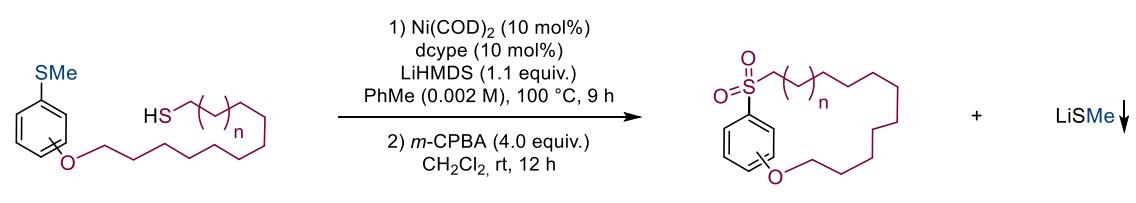
Parameters

- Concentration

- Catalyst loading

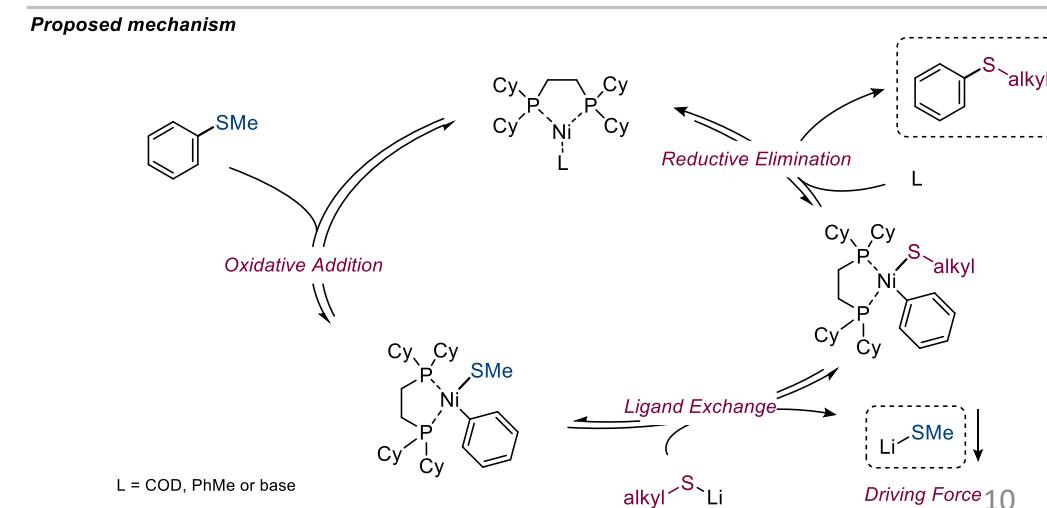
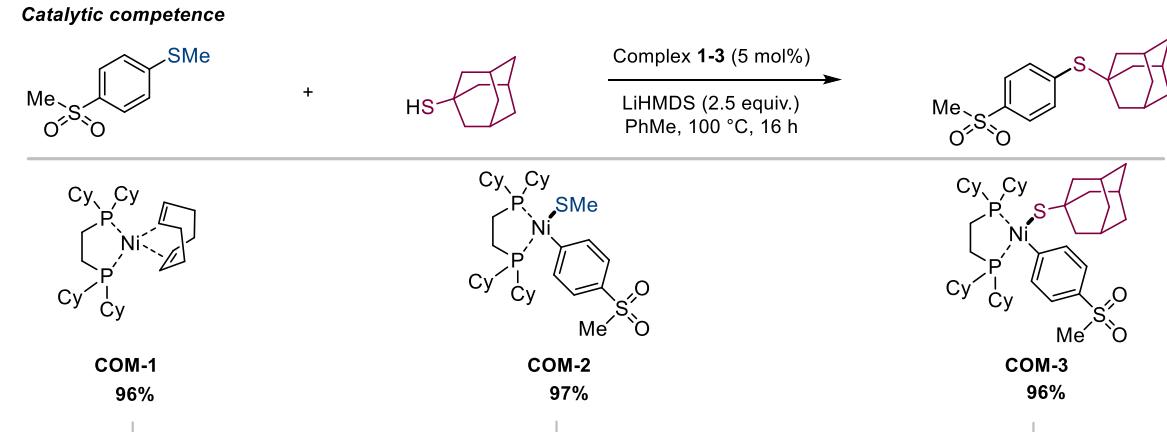
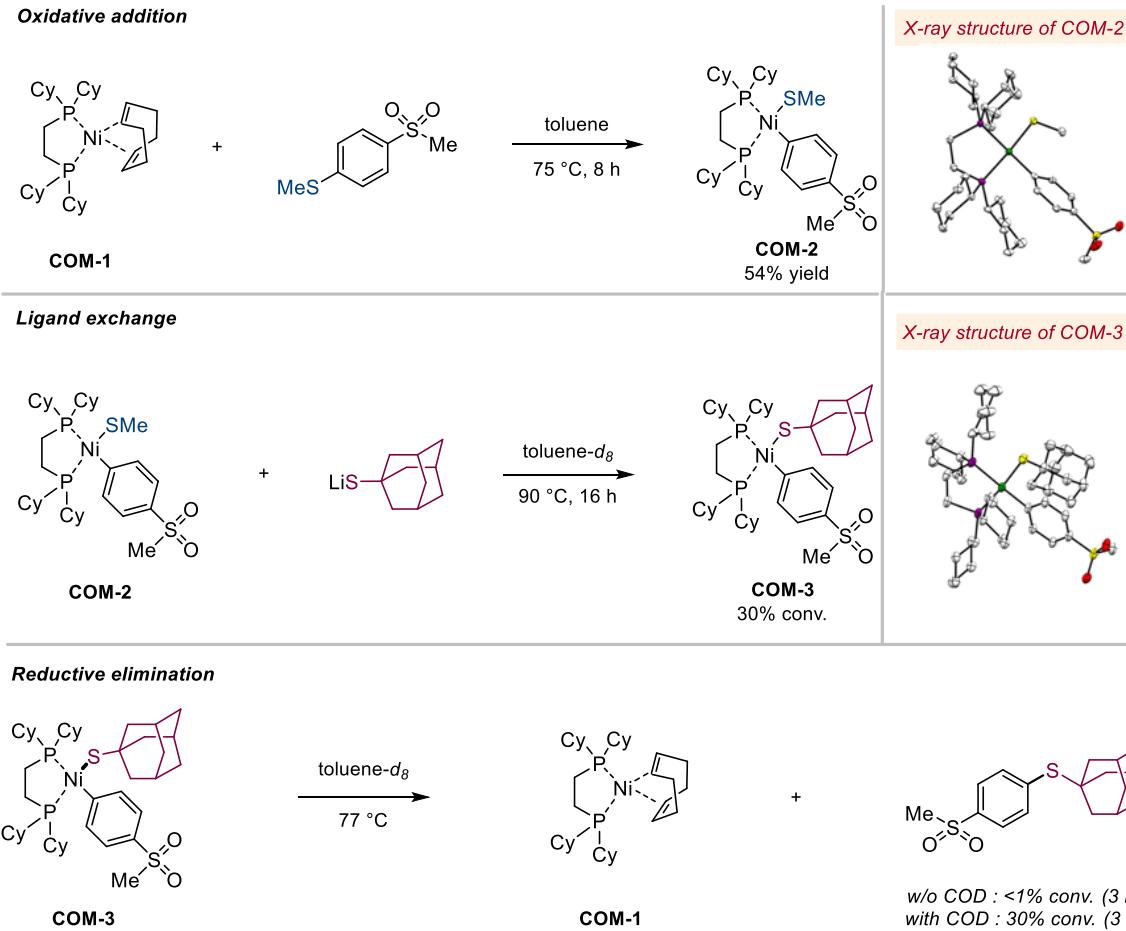
- Time

- Temperature

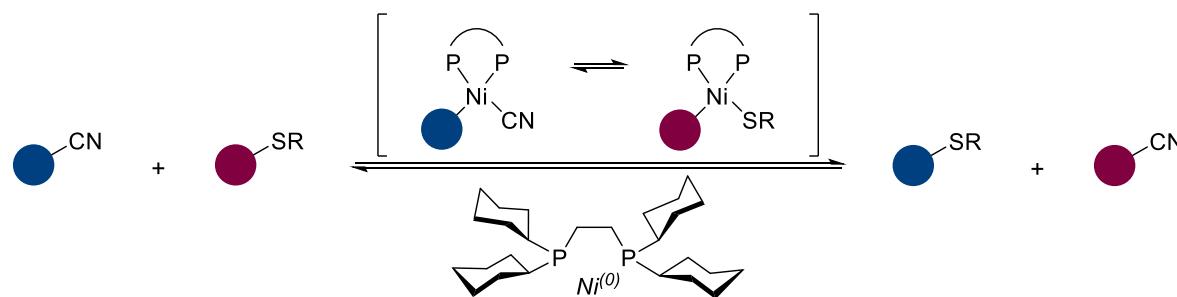


Reversible Inter- and Intramolecular Aryl Thioether Metathesis

Mechanistic study

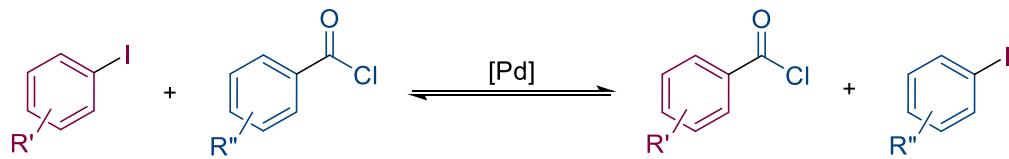


Reversible Functional Group Metathesis between ArCN and ArSR

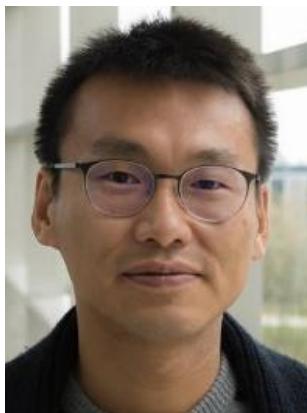


Reversible Functional Group Metathesis between ArCN and ArSR

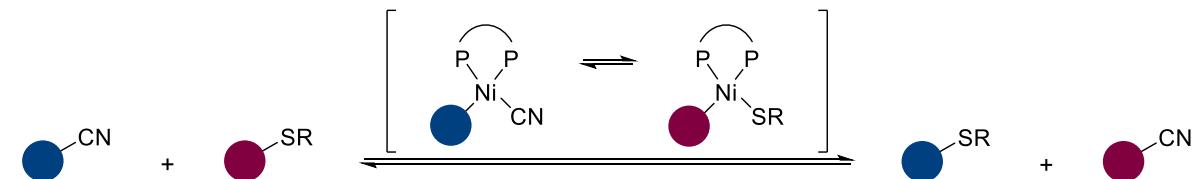
ArI - ArC(O)Cl metathesis



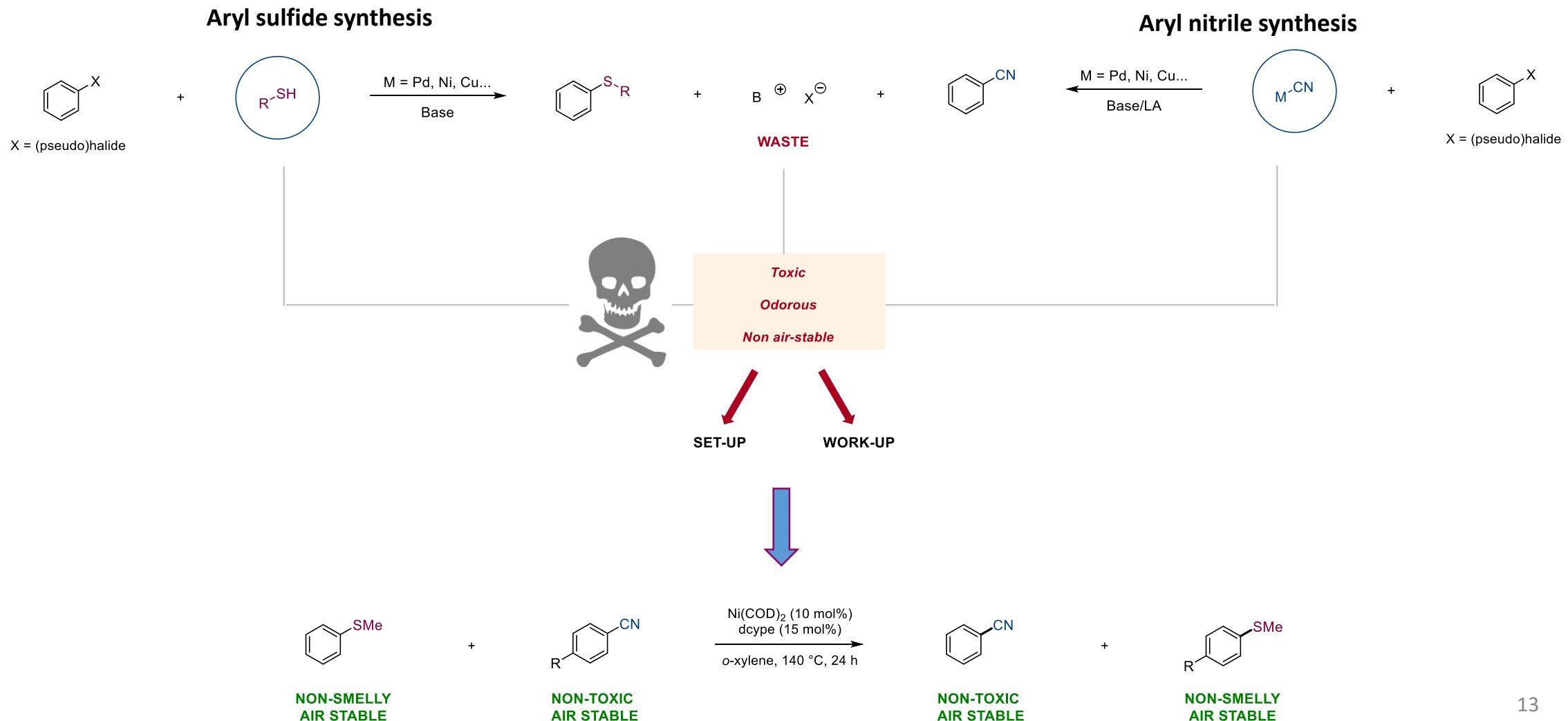
Nat. Chem. 2018, 10, 1016



Reaction design

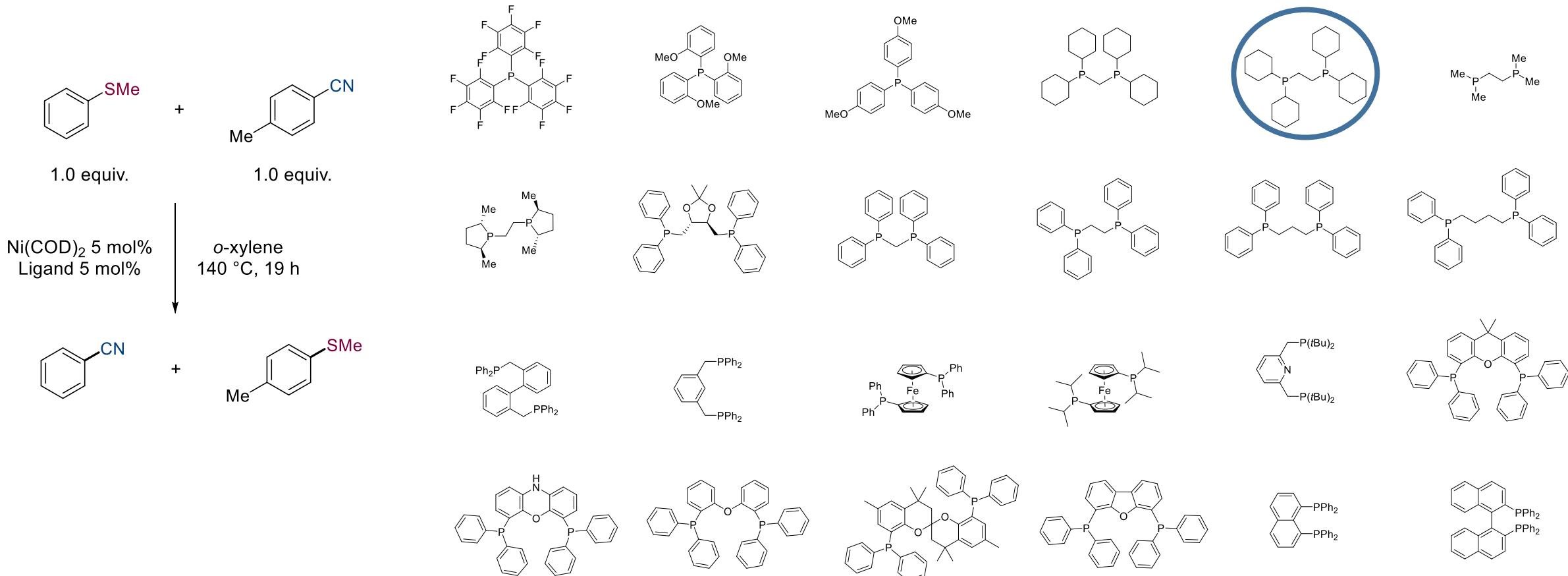


Reversible Functional Group Metathesis between ArCN and ArSR

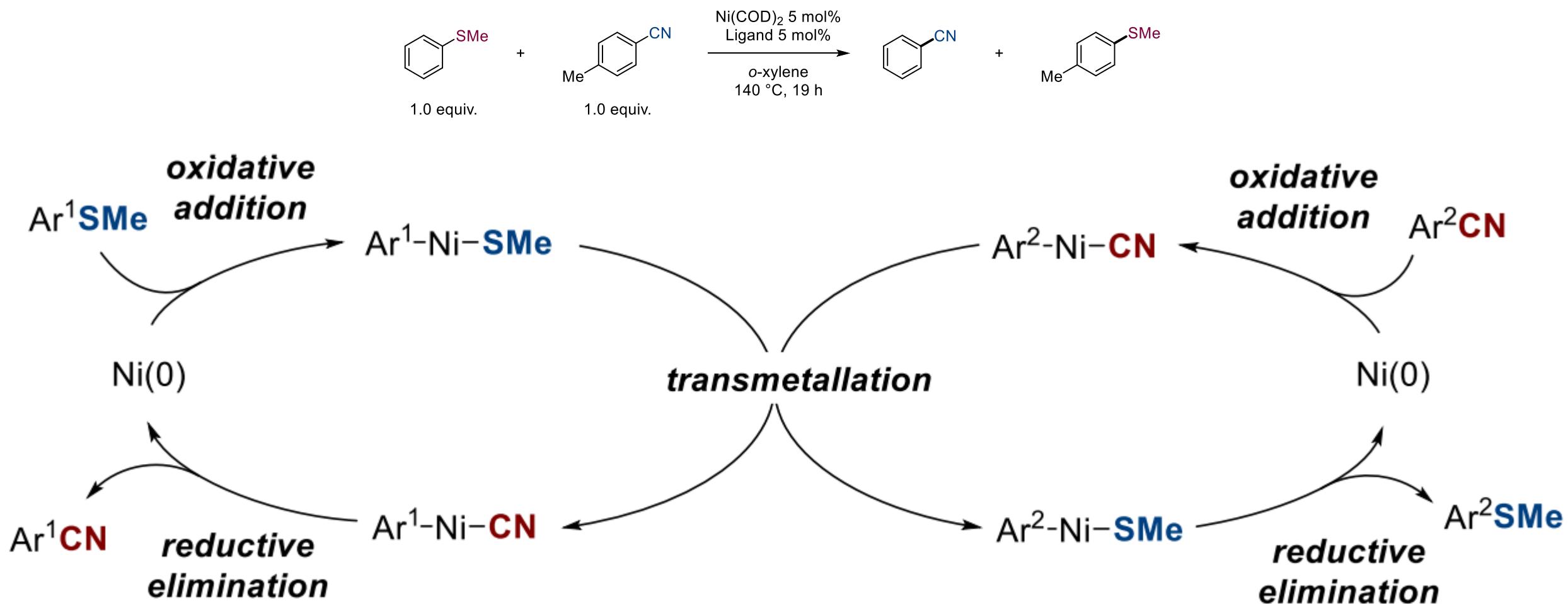


Reversible Functional Group Metathesis between ArCN and ArSR

Optimization

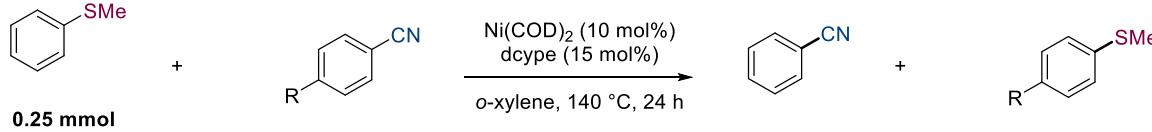


Reversible Functional Group Metathesis between ArCN and ArSR

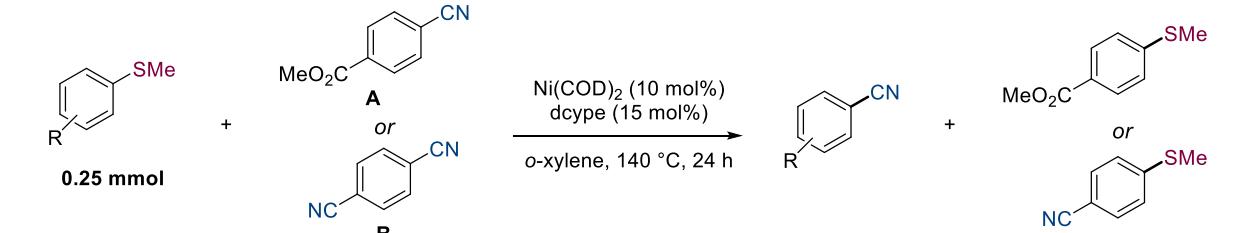


Reversible Functional Group Metathesis between ArCN and ArSR

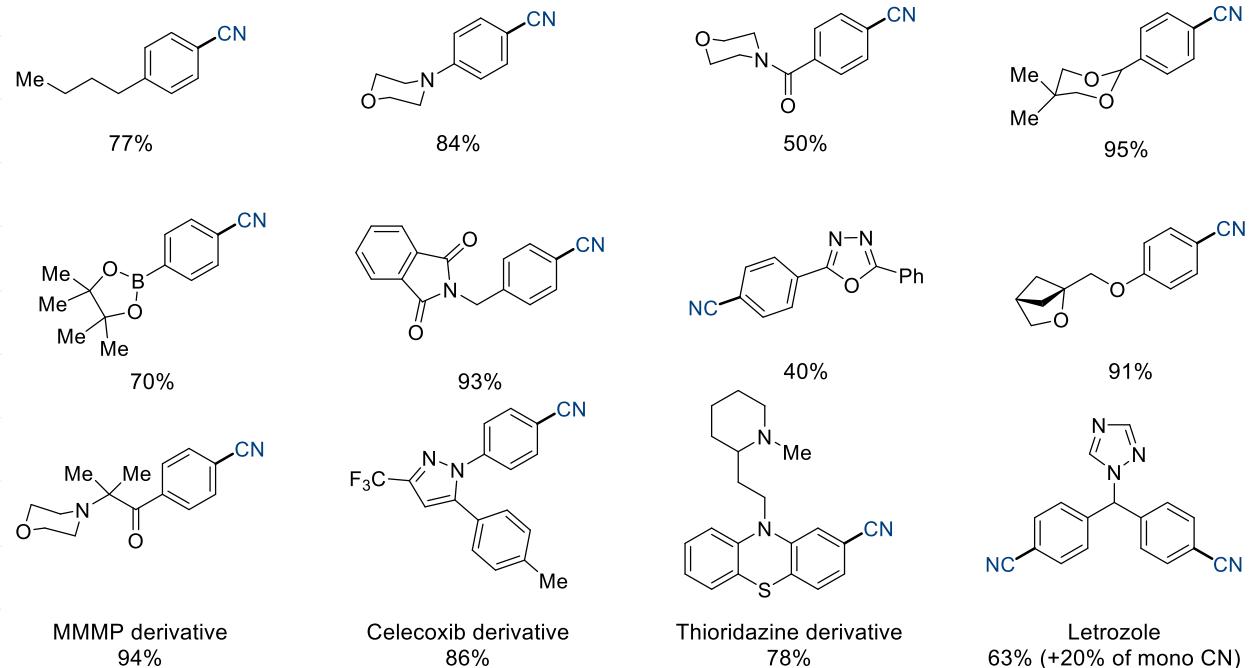
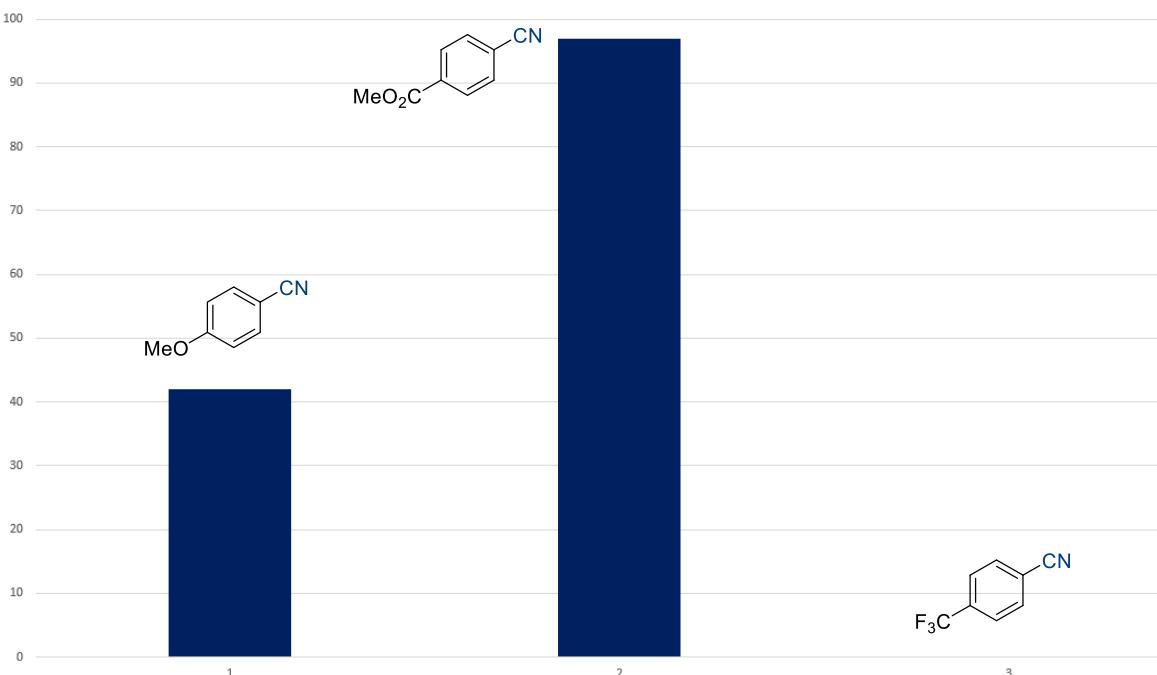
Driving the equilibrium



Scope Cyanation (Forward)

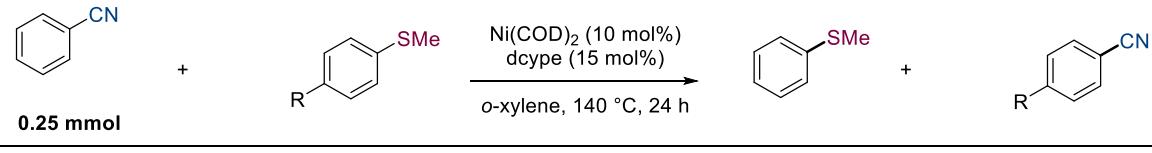


PhCN Yield (%)

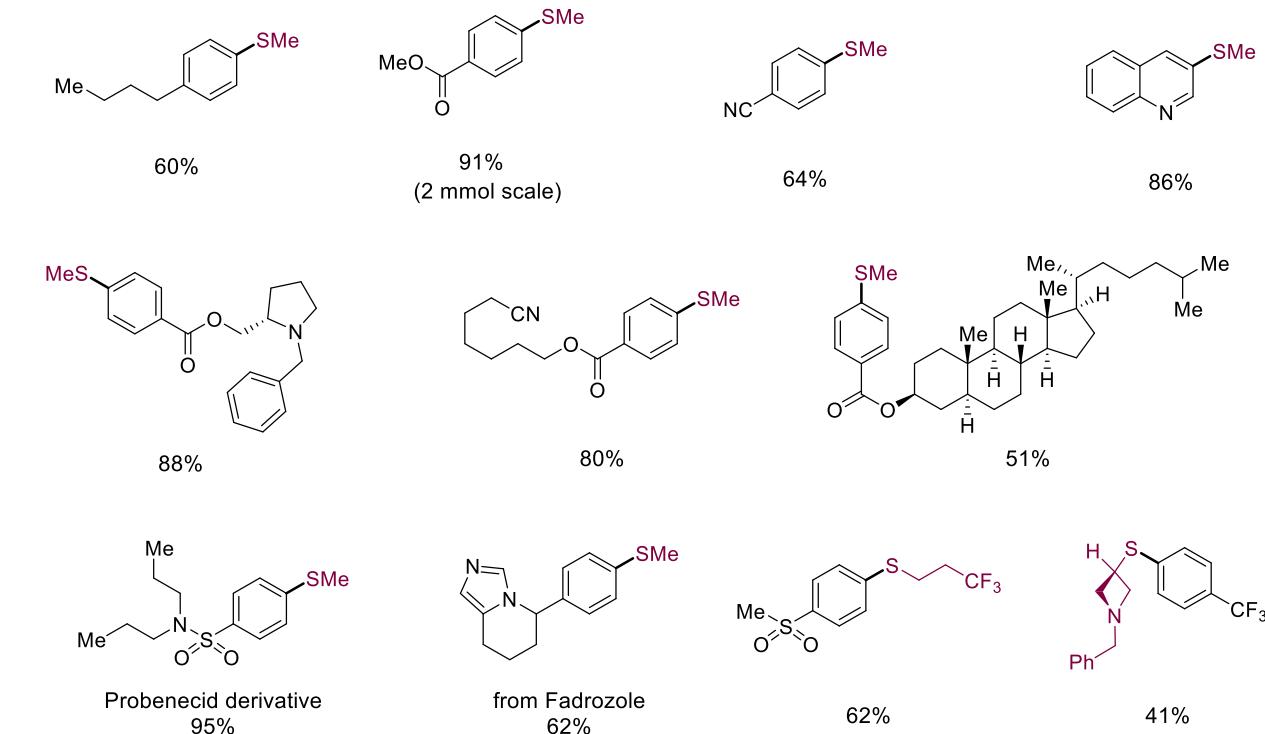
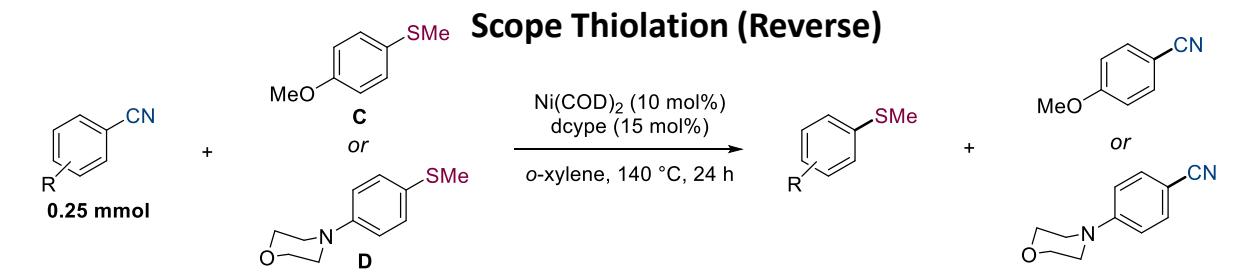


Reversible Functional Group Metathesis between ArCN and ArSR

Driving the equilibrium

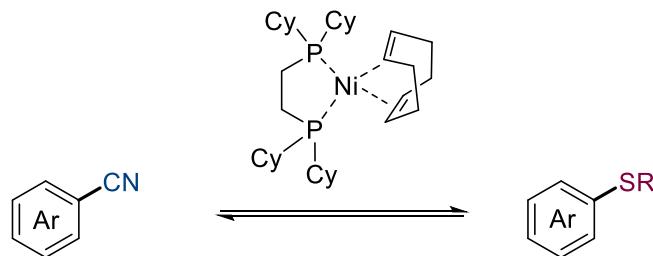


PhSMe Yield (%)



Reversible Functional Group Metathesis between ArCN and ArSR

Antagonist properties of CN & SR functionnal groups



Oxidant tolerant

Oxidant intolerant

Reductant intolerant

Reductant tolerant

Acid and base intolerant

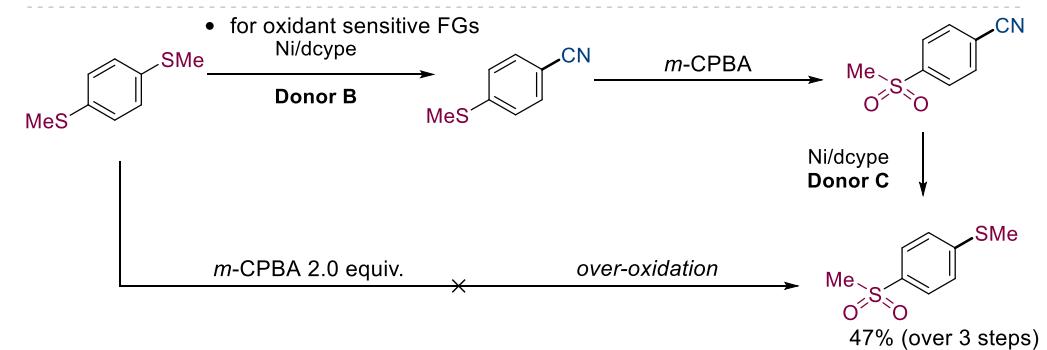
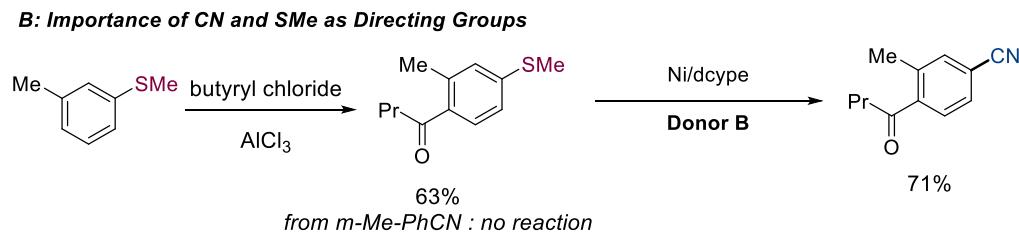
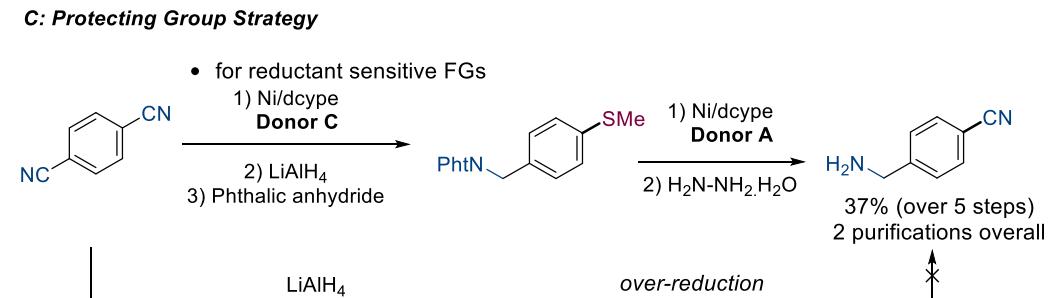
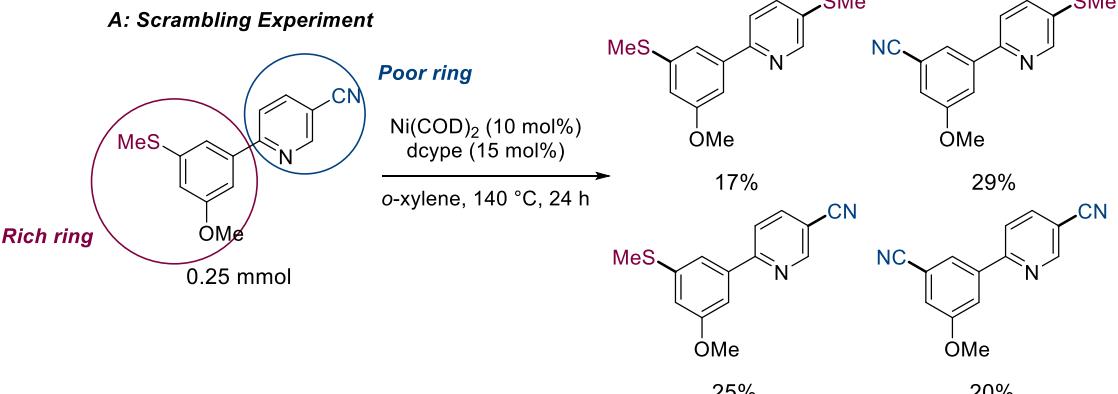
Acid and base tolerant

Electron-poor

Electron-rich

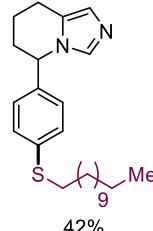
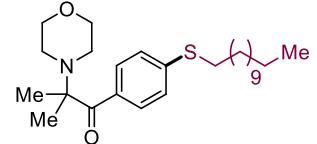
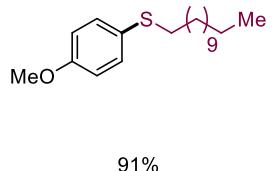
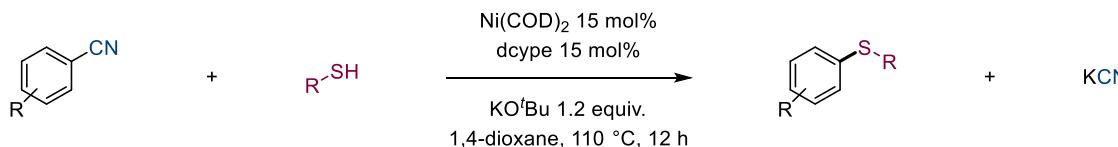
Reversible Functional Group Metathesis between ArCN and ArSR

Synthetic applications

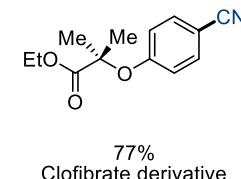
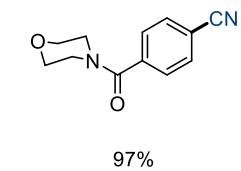
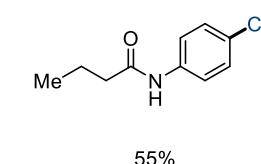
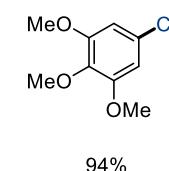
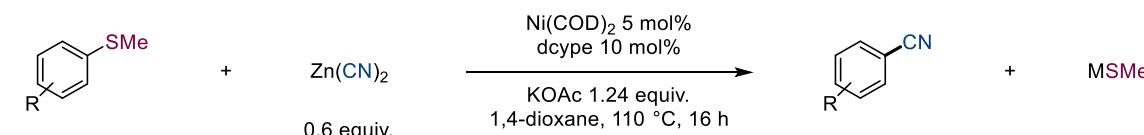


Reversible Functional Group Metathesis between ArCN and ArSR

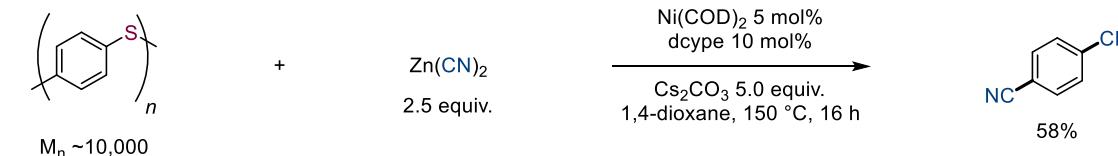
Thiolation with a thiol



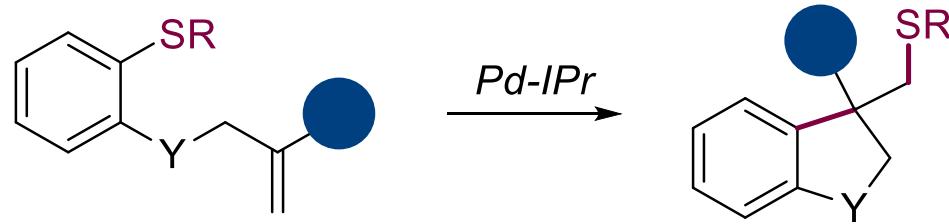
Cyanation with Zn(CN)_2



Depolymerization of a polymer

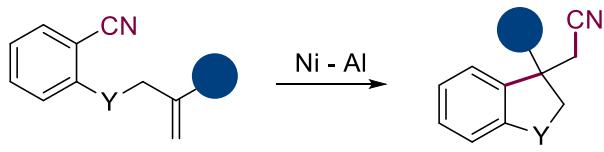


Carbothiolation of unsaturated of alkenes and alkynes

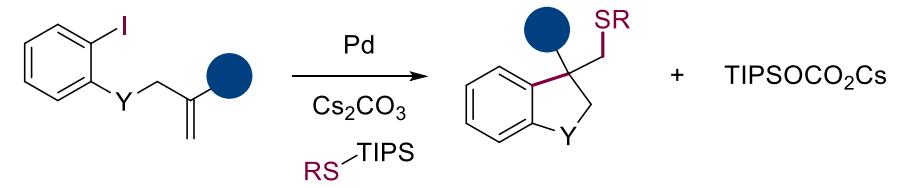


Carbothiolation of unsaturated alkenes and alkynes

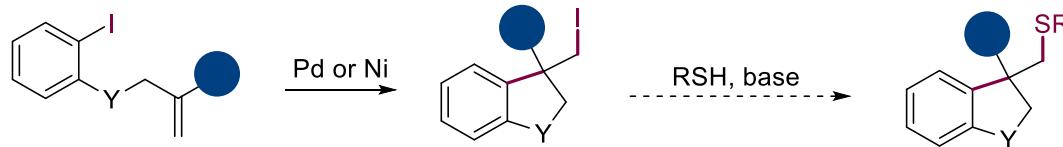
Background



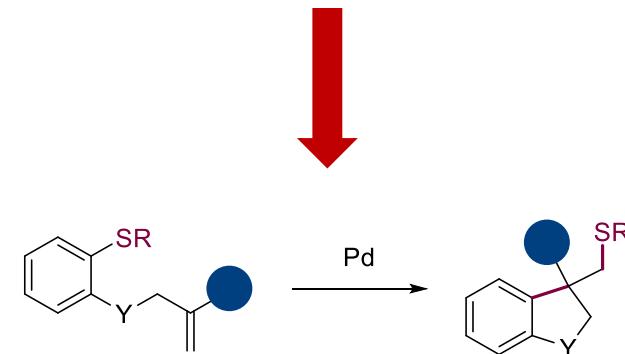
Hiyama, *J. Am. Chem. Soc.* **2008**, 130, 12874–12875
Jacobsen, *J. Am. Chem. Soc.* **2008**, 130, 12594–12595



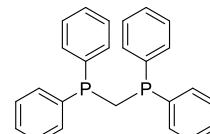
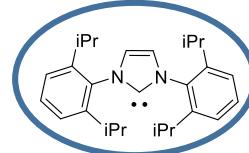
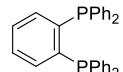
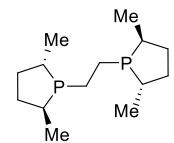
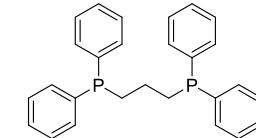
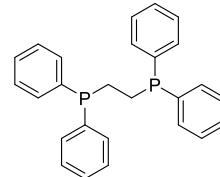
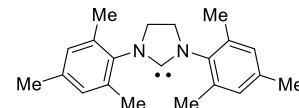
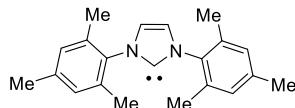
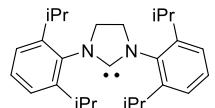
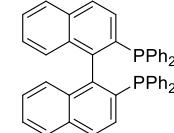
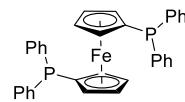
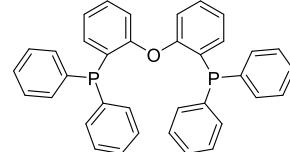
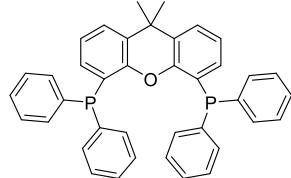
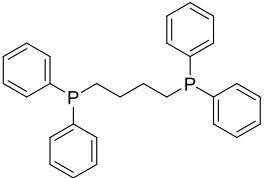
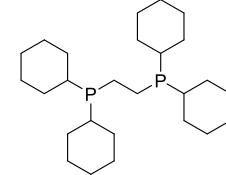
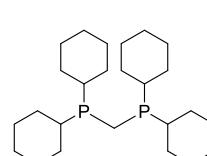
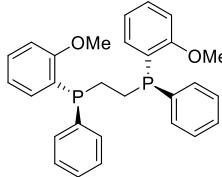
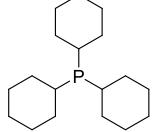
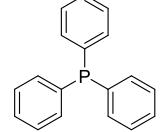
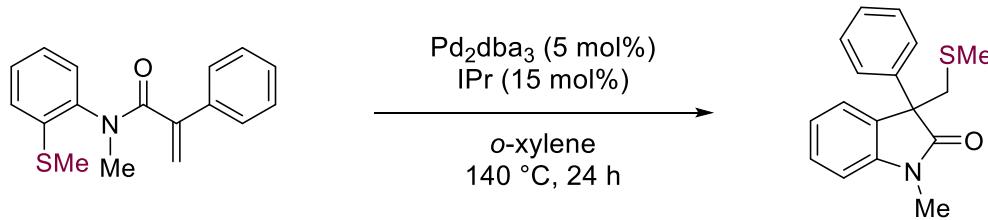
Nakada, *Org. Lett.* **2019**, 21, 8280–8284



Lautens, *J. Am. Chem. Soc.* **2011**, 133, 1778–1780
Tong, *J. Am. Chem. Soc.* **2011**, 133, 6187–6193



Carbothiolation



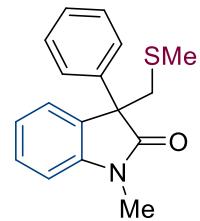
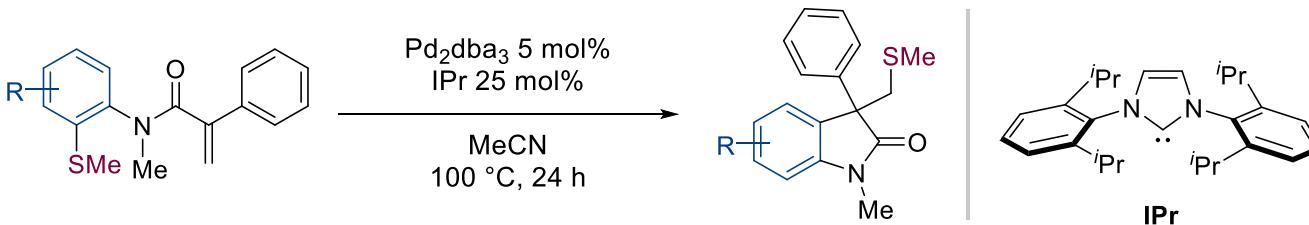
Further screening:

- Cat. Load.
- Pd source
- Solvent
- Temperature

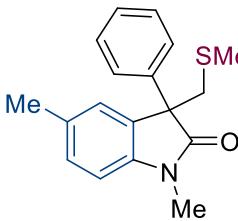


89% yield

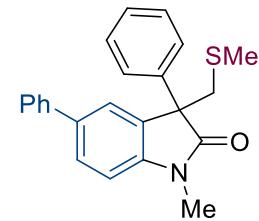
Carbothiolation



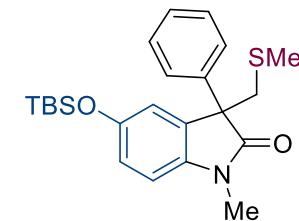
1, 89%
85% [5 mmol scale]



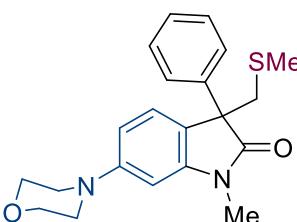
2, 90%^b



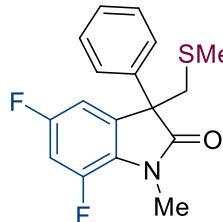
4, 97%



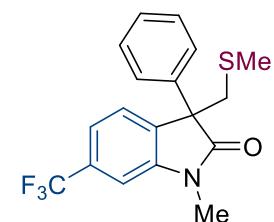
6, 84%



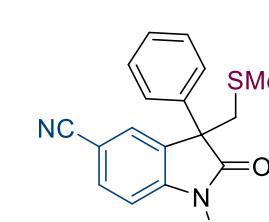
7, 81%



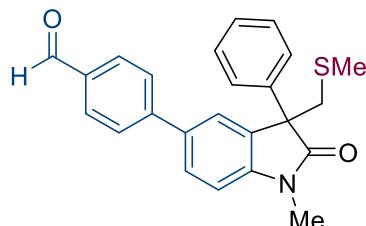
9, 87%



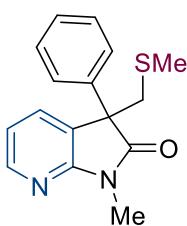
11, 92%



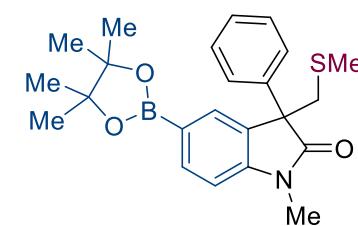
12, 94%



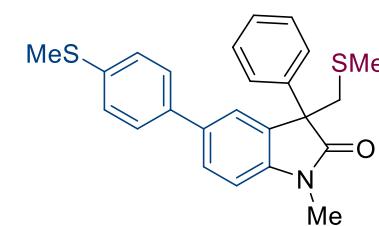
13, 51%



14, 60%

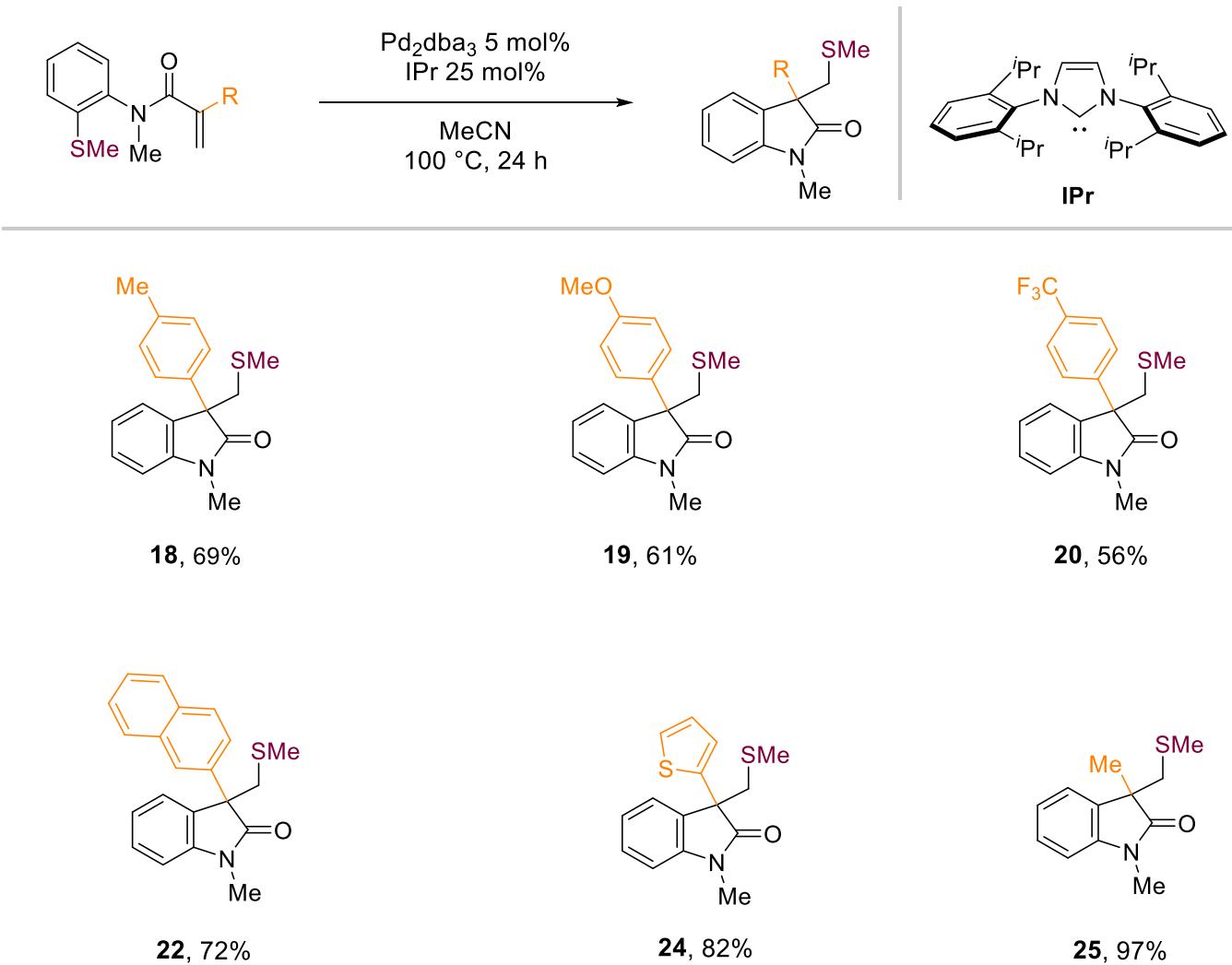


16, 93%

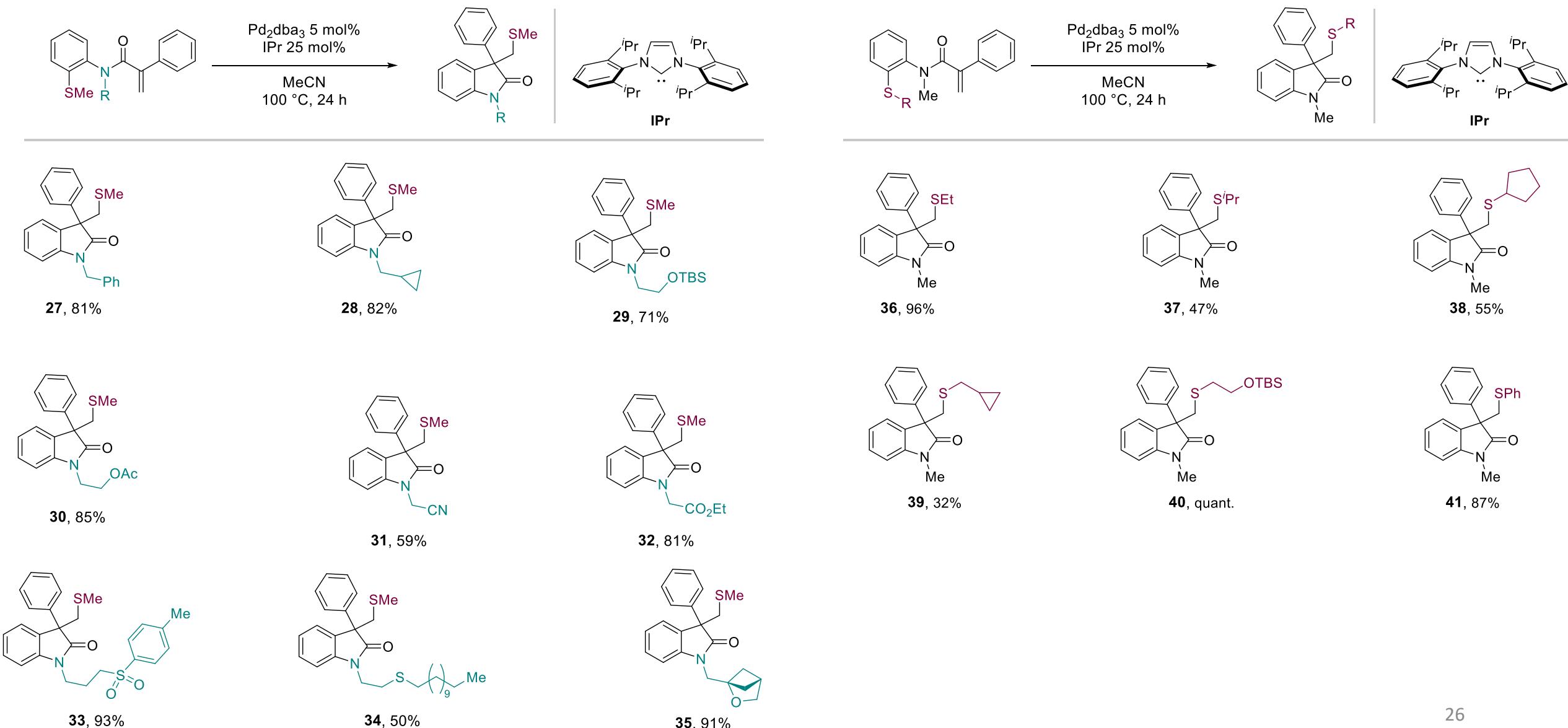


17, 89%

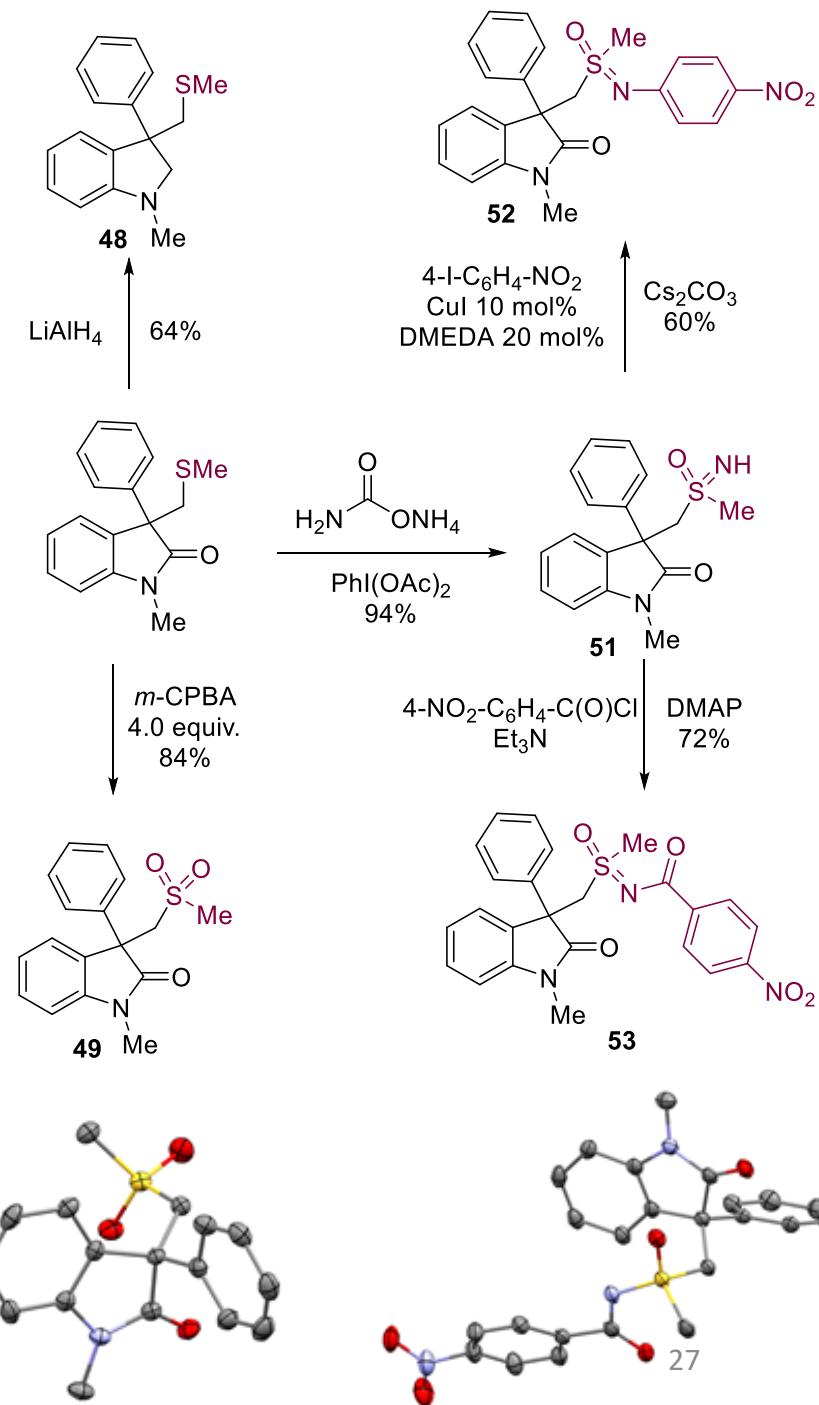
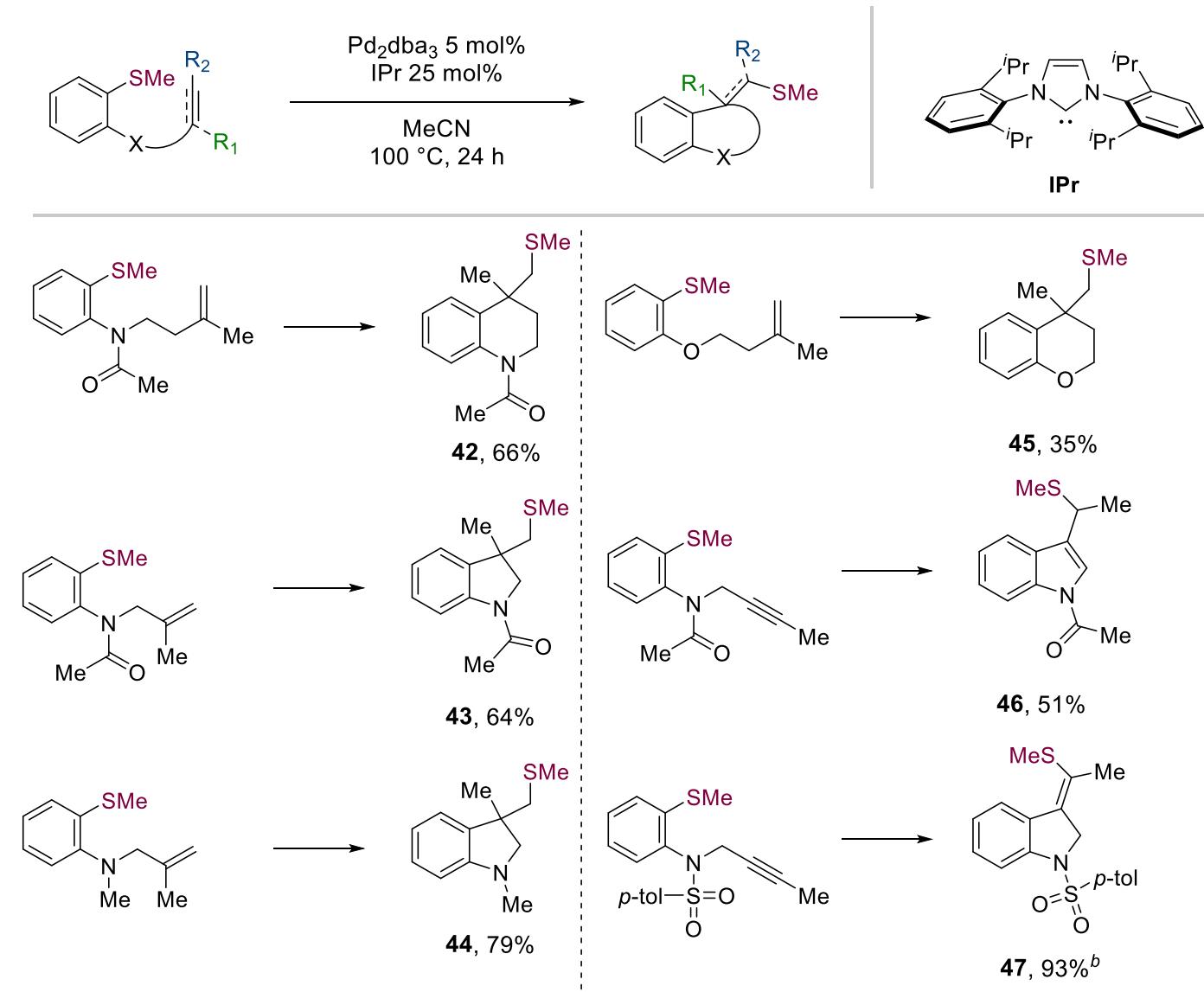
Carbothiolation



Carbothiolation



Carbothiolation

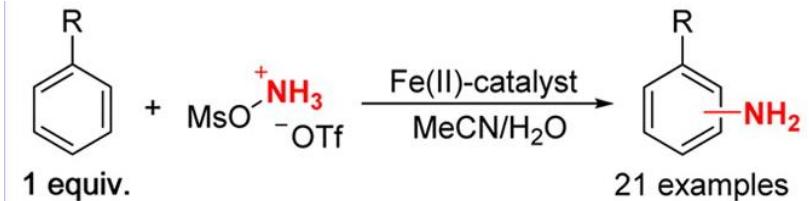


Other projects

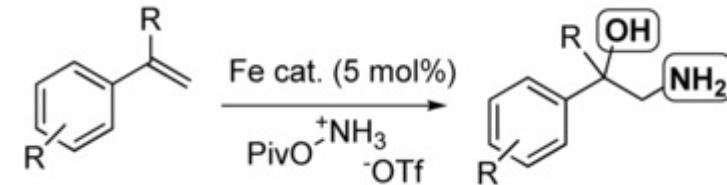
Amination of alkenes



||

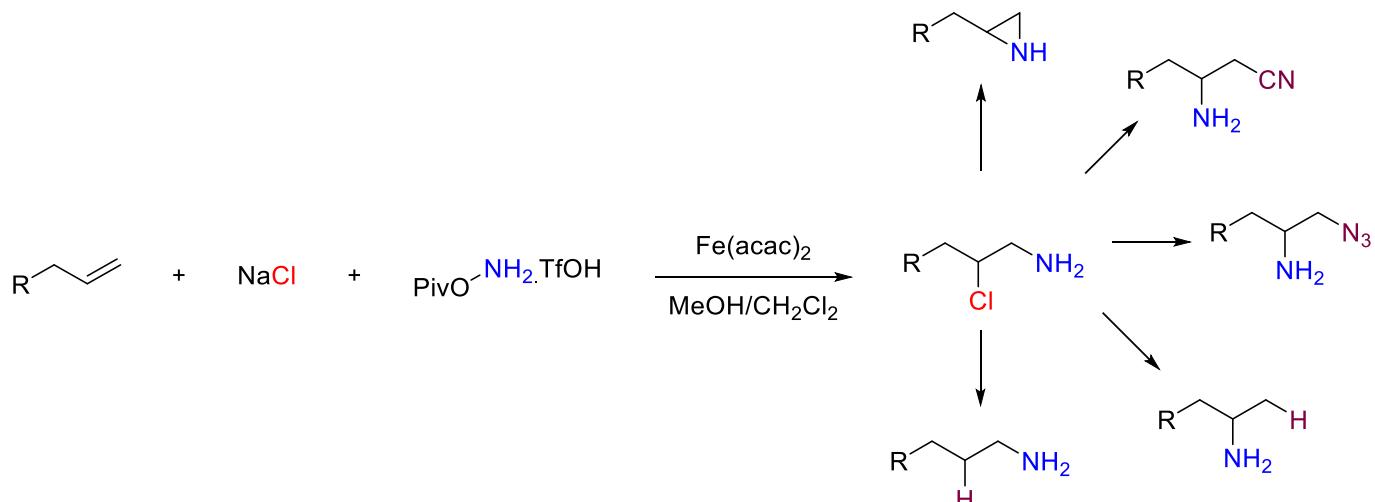


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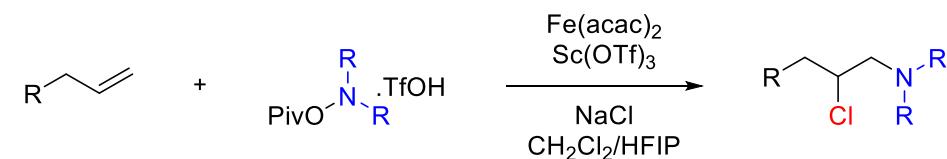


Angew. Chem. Int. Ed. 2016, 55, 2248-2251

Chloro-amination of alkenes



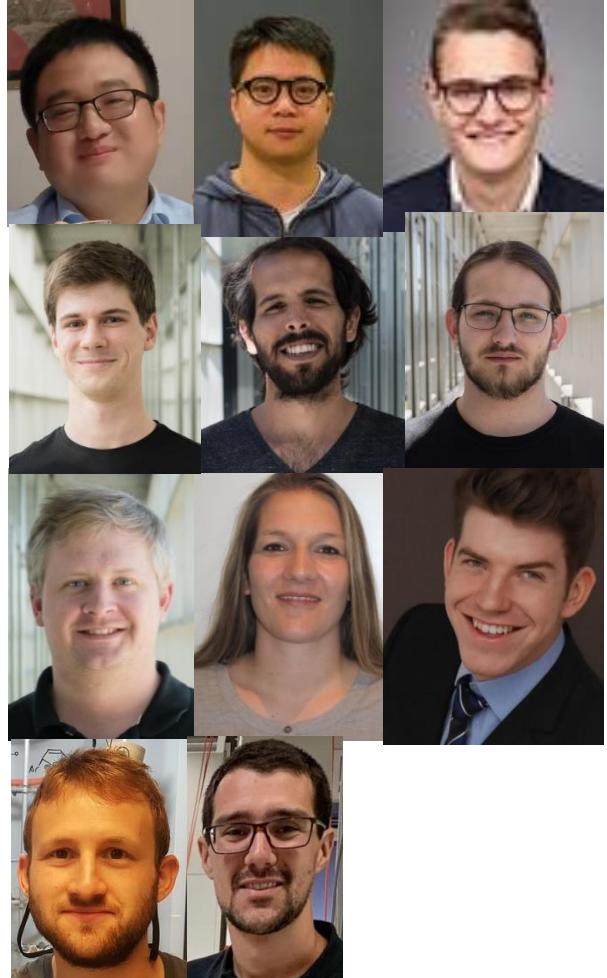
Chloro-alkyl amination of alkenes



Acknowledgments



Prof. Dr. Bill Morandi



and all former group members.