

## Radical Polycyclizations in the Synthesis of Heterocyclic Natural Products

Dan Forster

LSPN - Seminar

- Introduction
- Polycyclization cascades
  - C-centered radical initiated cascades
    - aldehydes
    - alkyl
    - alkenes
    - alkynes
  - N-centered radicals
    - N-aminyl radical
    - N-amidyl radical

#### Why radical polycyclizations?

Cascade polycyclization reactions => huge interest to shortly access complex frameworks

- Pericyclic reactions (electrocyclizations, cycloadditions, ..)
  - Very elegant, robust, complex structures can be obtained from a single transformation
  - Substrates must be carefully designed (matched/mismatched orbitals)
- Ionic cascades (cationic, anionic)
  - Present in biosynthetic pathways, nature's choice, well implemented
  - Difficult to control due to many possible rearrangements/shifts (carbocations), low functional group tolerance
- Radical cascades
  - Very reactive, extraordinarily short access to complex frameworks from simple starting materials



#### Radical generation

#### **Homolysis**

Heat or light

$$x - x \longrightarrow 2x$$

example:

$$\begin{array}{c|c}
CN & N_2 & CN \\
\hline
N_2 & 2 & \vdots \\
CN & CN & \vdots
\end{array}$$

$$R_0 O_R \longrightarrow 2 R^0$$

Homolytic cleavage of weak, non polarized bonds

# Single-Electron Reduction / C-H bond oxidation

Electron addition on polarized bonds or metal oxidation of activated C-H bonds

## Metal mediated hydrogen atom transfer

$$M_X H \longrightarrow H + M_{X+1}$$

M = Mn, Fe, Co, Sn

"Homolytic" cleavage of transition metal hydrides generating highly reactive hydrogen radical species

Once generated, radicals tend to propagate => well suited for cascade reactions to rapidly access complex frameworks.



#### Pioneering work: Porter (1970s): peroxide initiated cascades for prostaglandins synthesis

Observations in peroxy radicals cyclization, which were further used for the synthesis of prostaglandins analogues

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Prostaglandin Analogues

Porter, J. Am. Chem. Soc. 1975, 97, 1281; J. Org. Chem. 1975, 40, 3614

Pioneering work: Ueno-Stork 1980s: halo-abstraction and substrate design for regiocontrolled radical cascades

Bu<sub>3</sub>SnH

Pioneering work: Curran 1980s: Total synthesis of Hirsutene employing radical polycyclization

First radical polycyclization in the context of total synthesis

- Intro radical generation
- Polycyclization cascades
  - C-centered radical initiated cascades
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## **EPFL**

#### Sm(II) used for generation of ketyl radical via single electron transfer (SET)

Chelating effect of the ester on the Sm(III) induces:

- Selective aldehyde reduction
  - High diastereoselectivity

Procter, Chem. Eur. J. **2013**, 19, 6718 Initial discory of Sml<sub>2</sub> as one electron reducing agent: Kagan, J. Am. Chem. Soc. **1980**, 102, 2693–2698



#### Sm(II) used for generation of ketyl radical via single electron transfer (SET)

Chelating effect induces again high diastereoselectivity (single diastereomer)

Least hindered aldehyde reacts first

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#### Sn radical used for halo-abstraction to generate stable tertiary C-centered radical

limonoid skeleton

• First 6-endo occurs from top face of the pseudo-chair form due to sp<sup>3</sup> hybridized carbon atom next to double bond (C-OTBS) (C7) => stereodetermining step governed by hybridization of C7

Access to trans-decalin

Inoue, J. Org. Chem. 2021, 86, 6869



#### Sn radical used for halo-abstraction to generate stable tertiary C-centered radical

 First 6-endo occurs from bottom face of the pseudo-boat form due to sp<sup>2</sup> hybridized carbon atom next to double bond (C=O) (C9) => stereodetermining step governed by hybridization of C7

Notable access to thermodynamically non-favoured cis-decalin

Inoue, J. Org. Chem. 2021, 86, 6869



#### Sn radical used for halo-abstraction to generate stable tertiary C-centered radical

(±) puberuline C framework

MeO H H OMe
HO OMe
puberuline C

- First sythesis of the hexacyclic framework of Puberuline C
- Challenging 7-endo trig cyclization (DFT: optimal SOMO/LUMO interraction)

Inoue, Chem. Sci. 2016, 7, 4372

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#### Fe-mediated hydrogen radical generation and addition on alkene radical acceptor

Trans-decalin favored thanks to mismatched radical/acceptor philicity. Slow radical addition favours thermodynamic product.

Cis-decalin isomer observed due to steric repulsion with ester group.

Liu, Angew. Chem. Int. Ed. 2017, 56, 5849

Pattenden, J. Chem. Soc., Perkin Trans. 1, 2000, 3522

 $R_2 = Me/alkyl$ 

- Intro radical generation
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    - alkenes
    - alkyne
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#### An enyne radical cyclization triggering 1,5 HAT

Unexpected Tosyl deprotection in Barton – McCombie deoxygenation conditions triggered by 1,5 HAT during the synthesis of Dapholdhamine B

#### Sn radical addition on alkyne radical acceptor

Xu, J. Am. Chem. Soc. 2019, 141, 13043

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#### Ir photocatalytic single electron oxidation of N anion generated from basic deprotonation

First use of photocatalytic radical generation for natural product synthesis

Qin, Acc. Chem. Res. 2019, 52, 1877



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## Strychnos Alkaloids: (+)-strychnine

Qin, Org. Lett. 2019, 21, 252

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#### Sn radical used for benzoyl-abstraction to generate amidyl radical

Observed cyclization product when using hydrogen substituted precursors

The chlorine substituent is introduced to direct 6-endo versus 5-exo cyclization

Zard, Angew. Chem. Int. Ed. 2002, 41, 1783; Synlett, 1996, 12, 1148; Tetrahedron lett. 1995, 36, 8791; Tetrahedron Lett. 1999, 39, 2125

### Cascade radical polycyclization reactions

- Well implemented in nowadays total synthesis
  - Ultra short access to complex frameworks
  - Radicals propagate until being "stopped"

=> "infinite" potential but also very challenging cascade design

Thank you all for your attention!



Scheme 1. Retrosynthetic analysis of hispidanin A.

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