

Magnetism

Two fundamental parameters

1) magnetic moment μ or magnetization (order parameter)

Define the intensity of the magnetic field generated by the system, via $\underline{H} = -\text{grad}(\underline{\mu} \cdot \underline{r} / r^3)$ and the intensity of the force acting on the system if immersed in a magnetic field, via $\underline{F} = \text{grad}(\underline{\mu} \cdot \underline{H})$

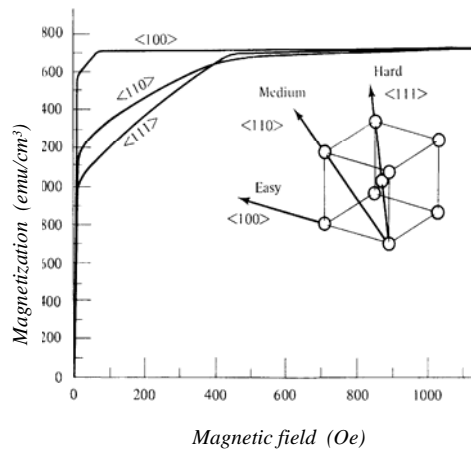
2) magnetic anisotropy energy K

Define the space direction/s along which the magnetic moment prefer to align

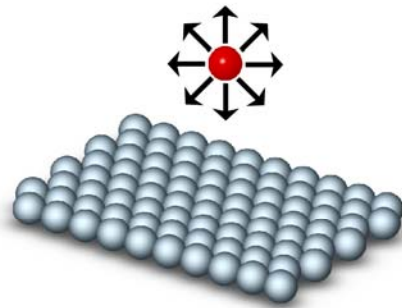
Fe bcc
easy axis: (100)

$$K_1 = 4.8 \times 10^4 \text{ J/m}^3$$

$$= 2.4 \text{ } \mu\text{eV/atom}$$



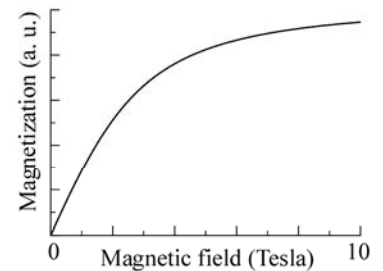
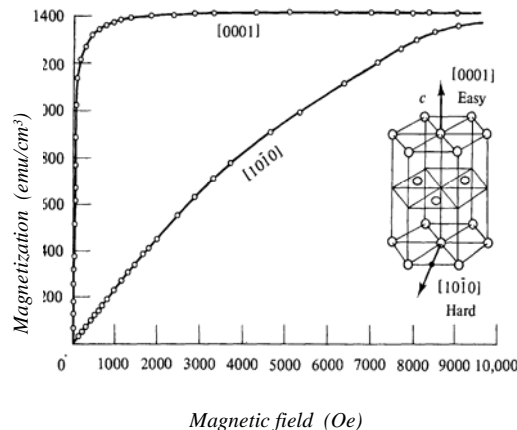
isotropic:
free magnetic atom



Co hcp
easy axis: (0001)

$$K_1 = 4.1 \times 10^5 \text{ J/m}^3$$

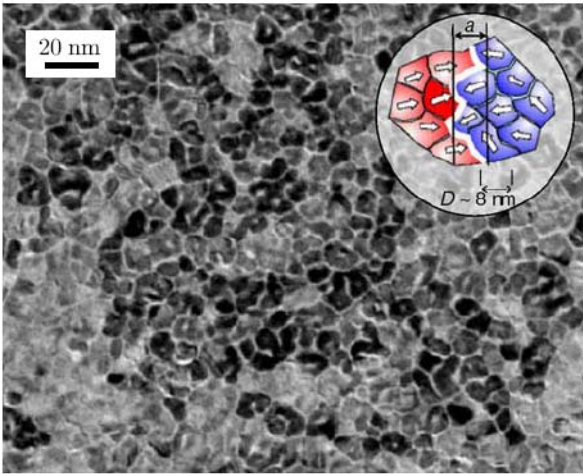
$$= 45 \text{ } \mu\text{eV/atom}$$



Complete isotropic
 $K = 0$

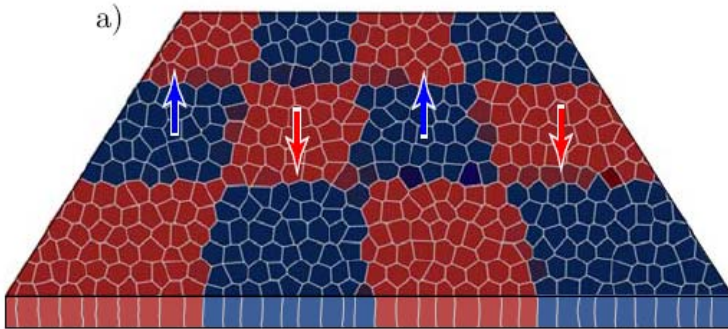
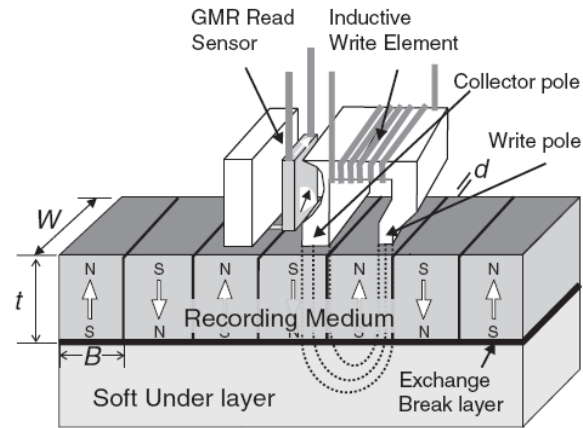
Bulk systems: K depends on the crystallographic structure

Information storage in magnetic memory



TEM images of the magnetic layer in a MRAM. Each bit is made of a few hundreds of grains

The bit size and shape is defined during writing by the head



Single particle bit



The dream

Atom in a magnetic field

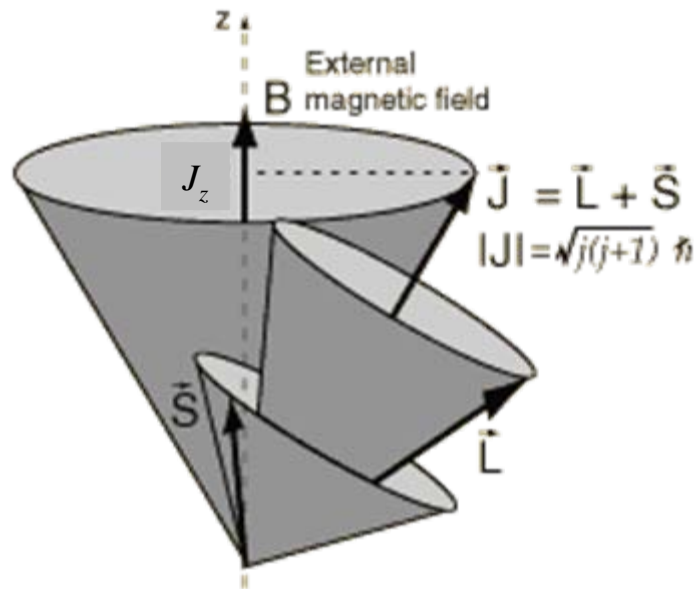
$$\mathcal{H} = \sum_i \left(\frac{\vec{p}_i^2}{2m} + V(\vec{r}_i) \right) + \frac{1}{2} \sum_{i \neq j} \frac{e^2}{|\vec{r}_i - \vec{r}_j|} + \mu_B B \frac{1}{\hbar} \sum_i (\ell_{z,i} + 2 s_{z,i})$$

Spin-orbit interaction:

Interaction of the spin of an electron with the magnetic field generated by its own orbital motion

$$\mathcal{H}_{ls} = \sum_i \zeta(\vec{r}_i) \vec{\ell}_i \vec{s}_i$$

$$\zeta(\vec{r}) = \frac{e\hbar^2}{2m^2c^2} \frac{1}{r} \frac{\partial V(r)}{\partial r}$$



Orbital magnetism: $m_L = L \mu_B$

Spin Magnetism: $m_S = g_e S \mu_B = 2 S \mu_B$

Atomic magnetic moment: $m_{at} = \mu_B(L + g_e S) = g J_z \mu_B = g J \mu_B$

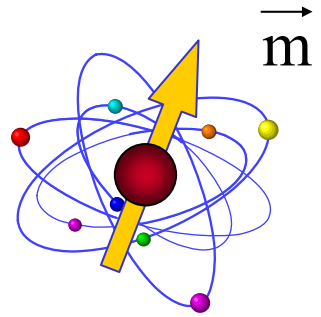
μ_B -> Bohr magneton

$g = 3/2 + [S(S+1) - L(L+1)] / (2J(J+1))$ is the Landé g-factor

Magnetism of an isolated atom

Magnetism is given by:

- 1) The spin moments of the electrons
- 2) The orbital moments of the electrons
- 3) The filling of the atomic orbital



Hund's rules:

- 1) Total spin $S = \sum_i s_i$ maximized
- 2) Total orbital momentum $L = \sum_i l_i$ maximized
- 3) L and S couple parallel ($J=|L+S|$) if band more than half filled
 L and S couple antiparallel ($J=|L-S|$) if band less than half filled

Ground state of a $3d^7$ ion (Co^{2+})

+2	+1	0	-1	-2
↑ ↓	↑ ↓	↑	↑	↑

$$L = 3, S = 3/2, J = 9/2$$

$$m_L = L \mu_B = 3 \mu_B,$$

$$m_S = g_e S \mu_B = 3 \mu_B,$$

$$m_{\text{at}} = g J \mu_B = 6 \mu_B$$

Ground state of a $3d^5$ ion (Mn^{2+})

+2	+1	0	-1	-2
↑	↑	↑	↑	↑

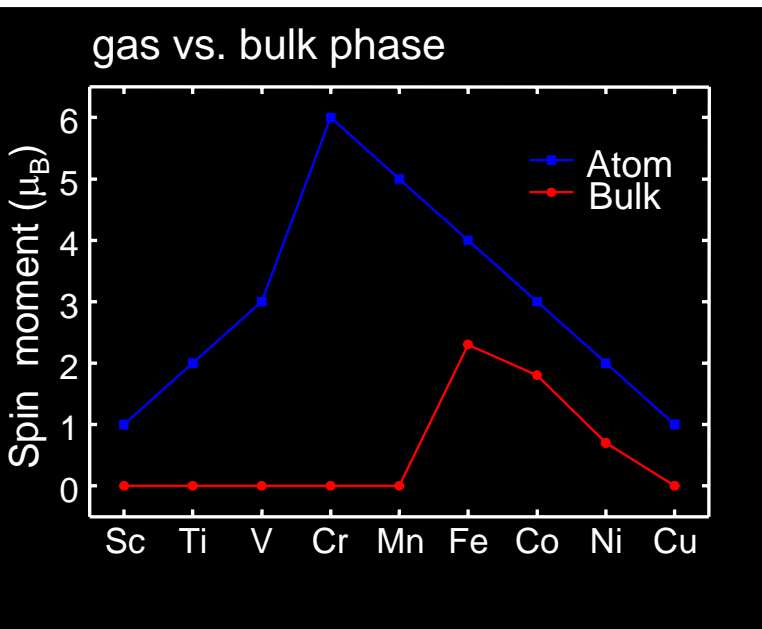
$$L = 0, S = 5/2, J = 5/2$$

$$m_L = L \mu_B = 0 \mu_B,$$

$$m_S = g_e S \mu_B = 5 \mu_B,$$

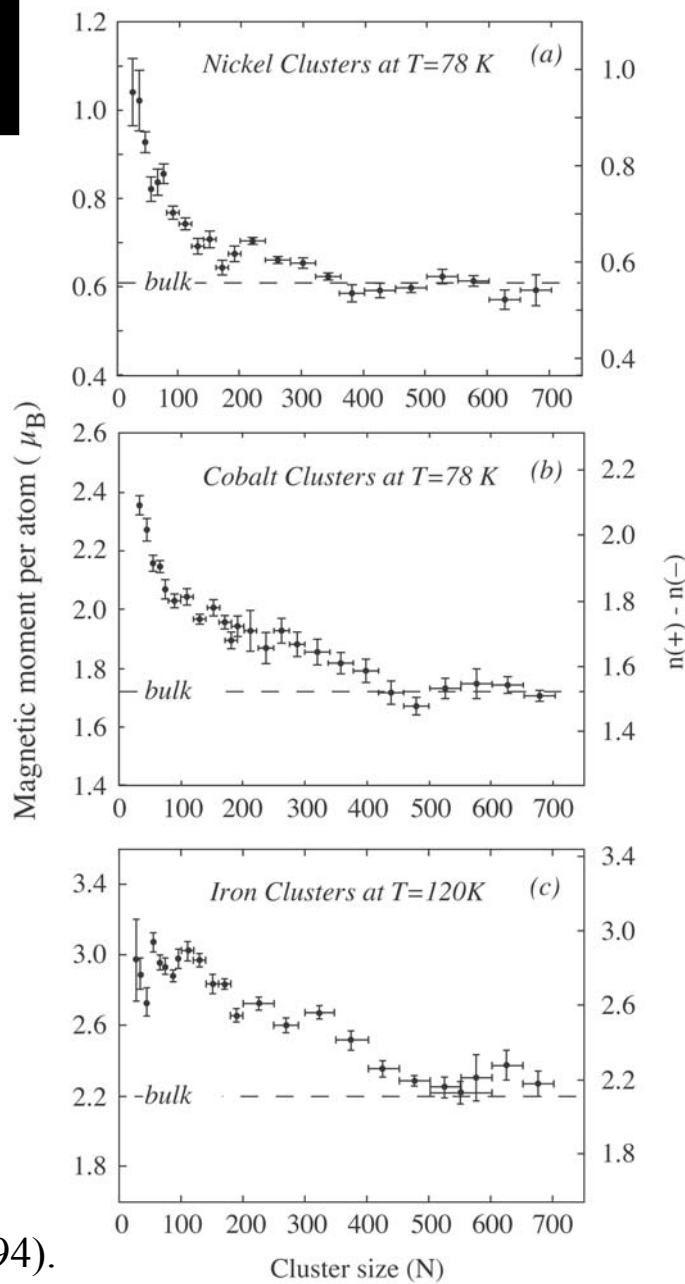
$$m_{\text{at}} = g J \mu_B = 5 \mu_B$$

3d elements: gas and condensed matter phase



mass-selected clusters in gas phase

$N > 20$ atoms



Bulk magnetism

Most atoms lose their magnetic moment when incorporated into a crystal (there are only a few magnetic solids). The main reason is the delocalization of the electrons due to the overlap of the wave functions of neighboring atoms.

Quenching of the orbital magnetic moment in 3d metals:

Experiments show that 3d ions in solids have $L \approx 0$, $J \approx S$ and $g = 2$.

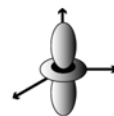
The crystal electric field produced by surrounding ions in a solid defines a particular set of (real) wave-functions for which the mean value of the orbital moment is zero (balance of $\pm m_L$).

d wave-functions in cubic symmetry

t_{2g}



$$d_{x^2-y^2} = 1/\sqrt{2} (|+2\rangle + |-2\rangle)$$

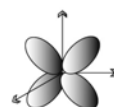


$$d_{3z^2-r^2} = |0\rangle$$

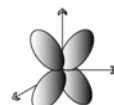
e_g



$$d_{xy} = i/\sqrt{2} (|+2\rangle - |-2\rangle)$$



$$d_{yz} = i/\sqrt{2} (|+1\rangle + |-1\rangle)$$



$$d_{xz} = 1/\sqrt{2} (|-1\rangle - |+1\rangle)$$

The crystal field gives a reduction of the orbital moment depending on its symmetry and strength.

Localized magnetism -> 4f and 5f states of the rare earth

Itinerant magnetism -> 3d (4d, 5d) states of transition metals

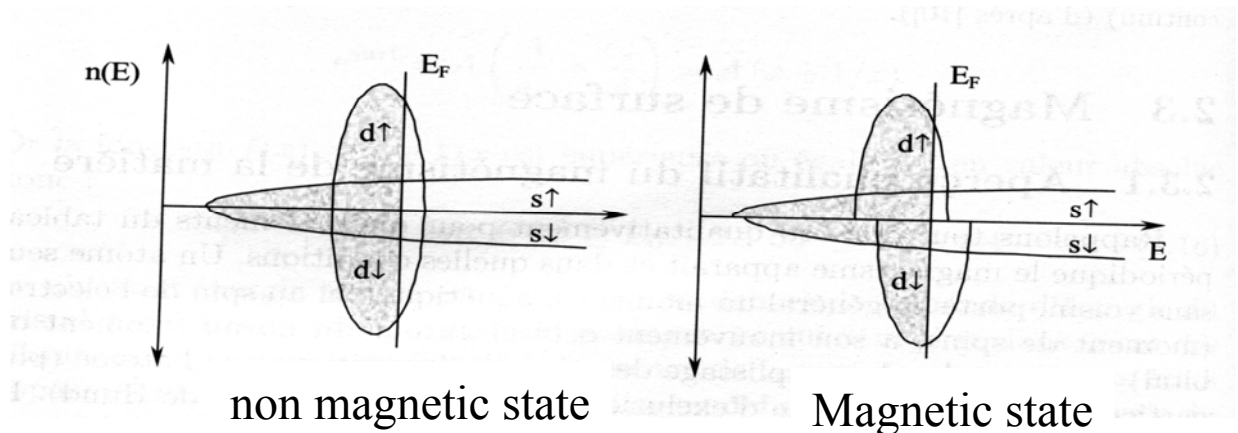
Itinerant magnetism

Itinerant magnetism -> delocalized electrons -> electronic bands

Stoner Exchange interaction:

Is the energy necessary to reverse the spin of one electron in the sea of all the other electrons -> spin up - down bands are shifted by $\delta = 1-2$ eV

$$\mathcal{H}_S = -1/2 \sum_{M,N} J_{M,N} \vec{S}_M \cdot \vec{S}_N,$$



Non magnetic state -> $n\uparrow = n\downarrow$

magnetic state -> splitting of electronic bands corresponding to opposite spin direction -> $n\uparrow \neq n\downarrow$ (δ band shift)

Spontaneous magnet -> $\Delta E \approx n(E_F) \delta^2 f(J) < 0$

Depends on $n(E_F)$, δ , and J

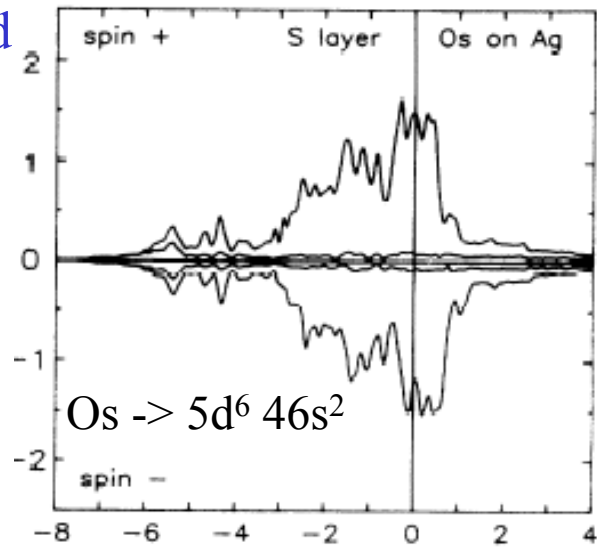
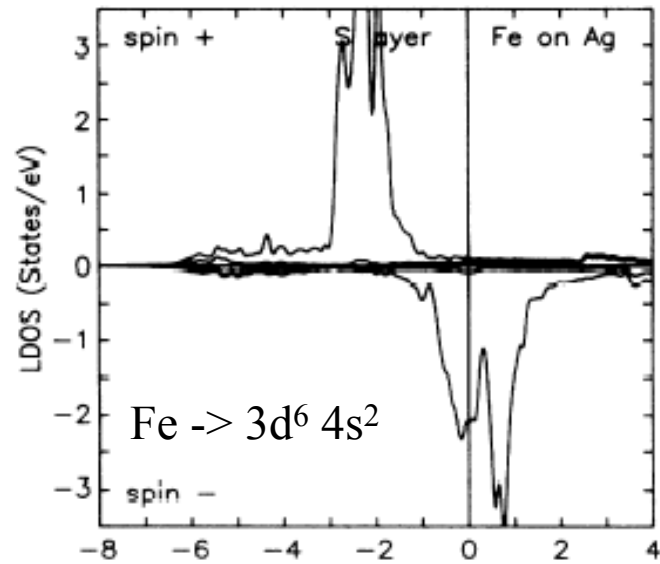
s bands are extended -> do not affect magnetism

d bands are localized -> their splitting determines the magnetism

Fe is magnetic

Os is non-magnetic

d-band



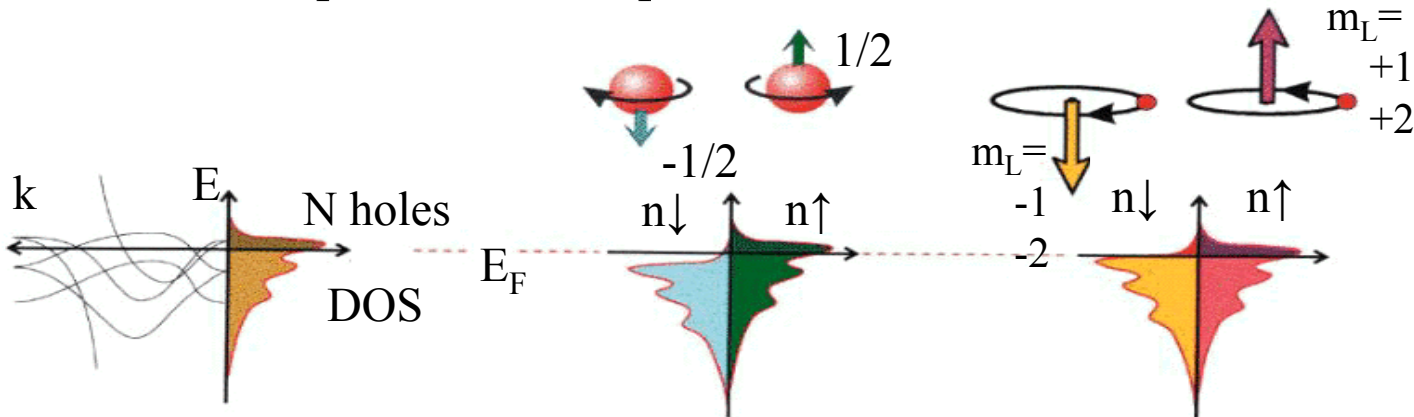
S. Blugel Phys. Rev. Lett. **68**, 851 (1992)

Orbital and spin magnetic moments in bulk magnets

d-band occupation

Spin moment

Orbital moment



Material	N holes
Fe	3.4
Co	2.5
Ni	1.5

m_s^{tot}	m_s^d	m_s^{sp}
2.19	2.26	-0.07
1.57	1.64	-0.07
0.62	0.64	-0.02

m_{orb}
0.09
0.14
0.07

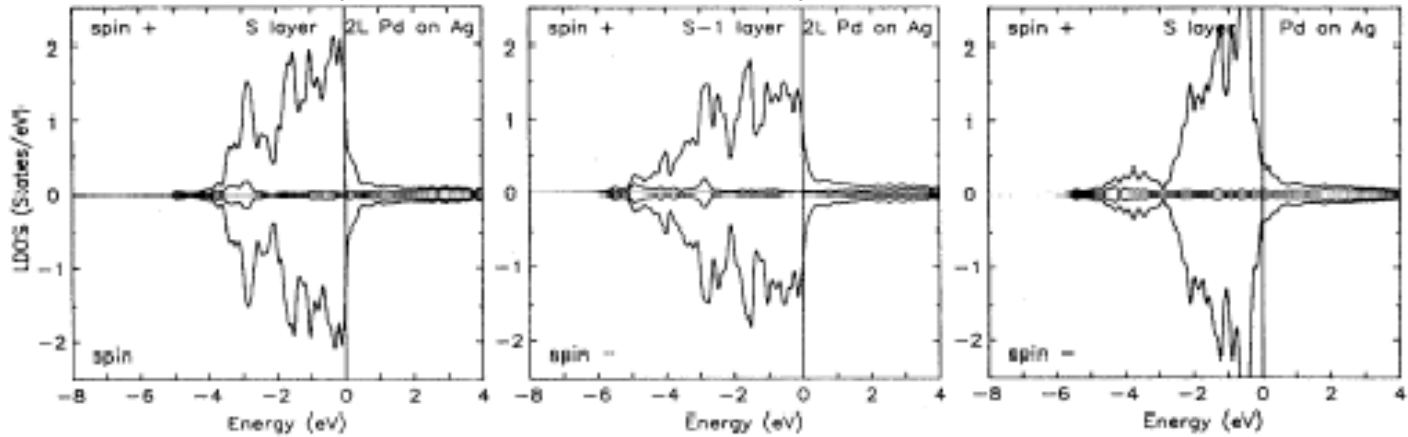
Surface magnetism

Narrowing of the electronic band due to the broken symmetry at surface

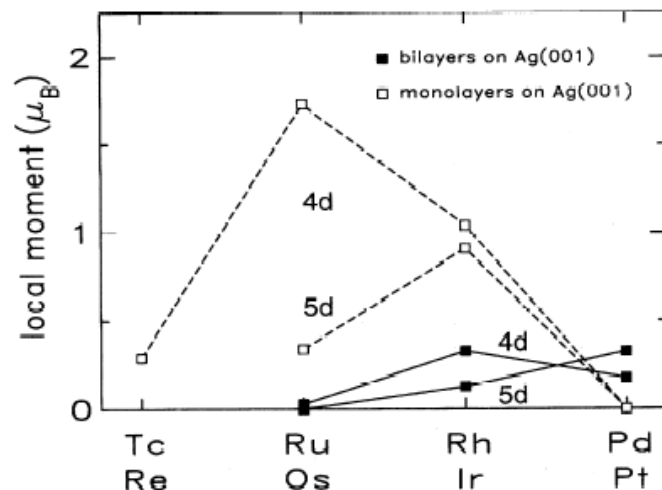
2 ML Pd on Ag(001)
Surface layer

2 ML Pd on Ag(001)
Interface layer

1 ML Pd on Ag(001)

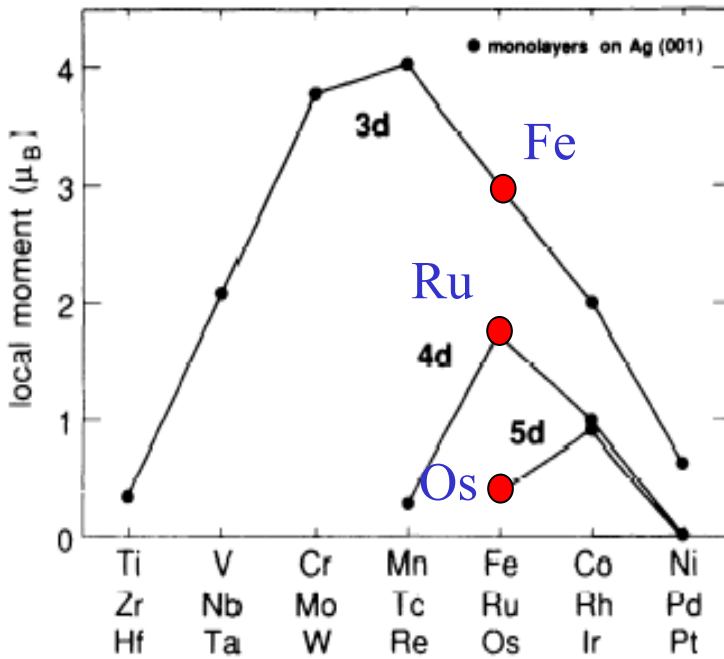
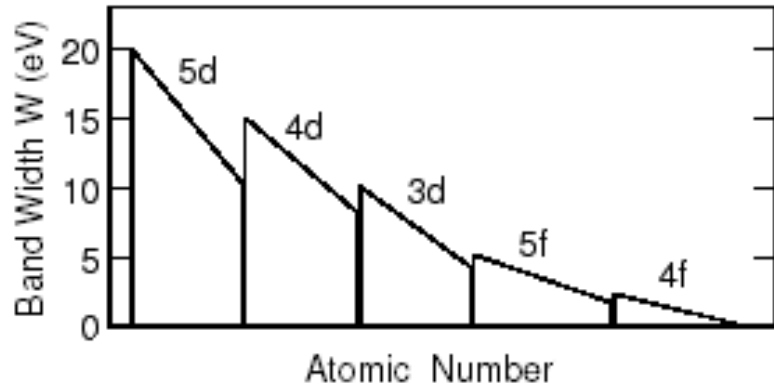


- 1) The Pd bi-layer bandwidth is wider than that of the mono-layer
Mean coordination in the bi-layer is larger than in the monolayer
- 2) Non magnetic elements in bulk state can become magnetic: Ex. Pd, Pt
- 3) The LDOS at the Fermi energy decreases moving from left to right
 - a) The magnetic moment of the atoms in the surface layer of the 2 ML Pd film have a larger moment than the atoms in the interface layer
 - b) The magnetic moment is larger in the 2 ML film than in the 1 ML film

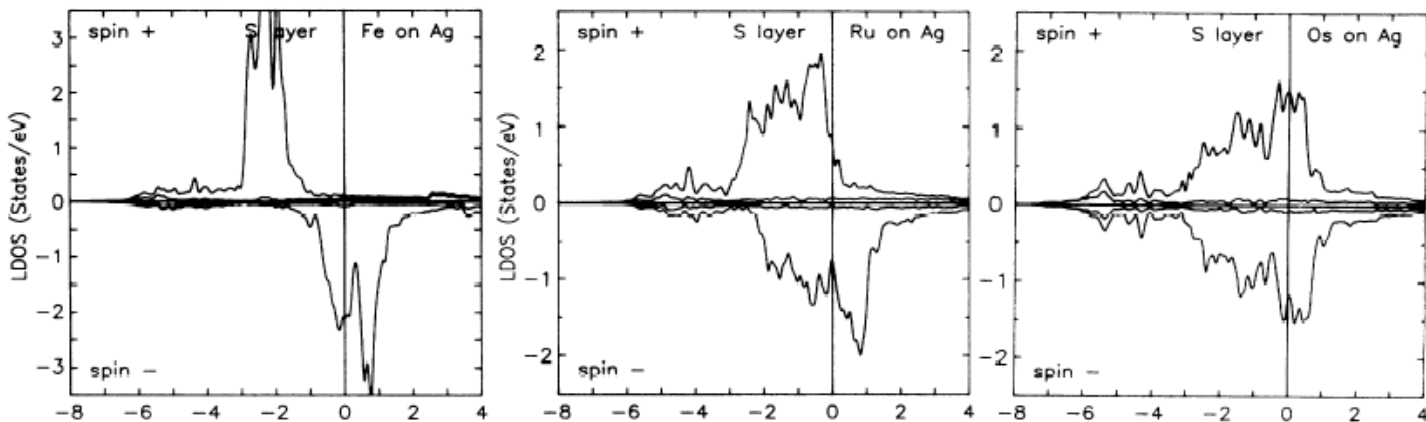


The electronic density at the Fermi level $n(E_F)$ increases moving from 5d to 4f elements

$$n(E_F) \sim 1/W$$

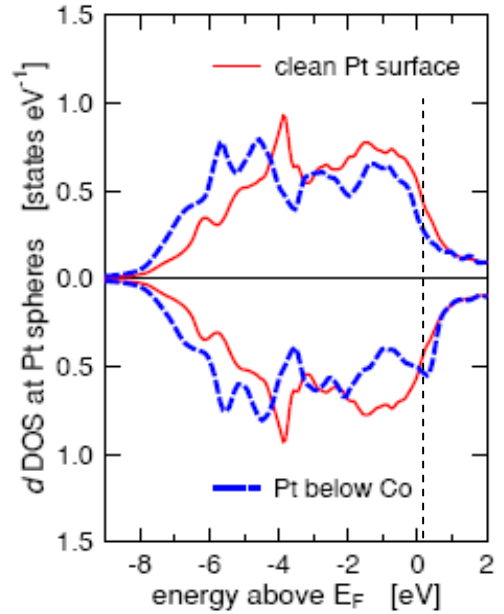
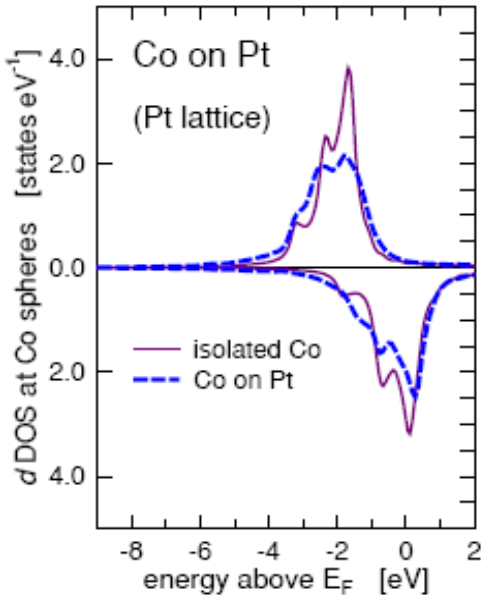


Monolayer on Ag(001)

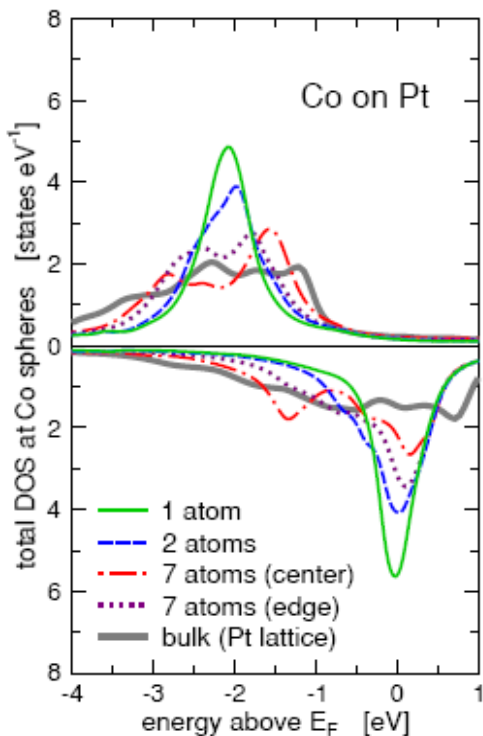


Co clusters on Pt(111)

Co monomer



Note: asymmetry in the Pt LDOS at E_F when covered by Co → induced magnetic moment

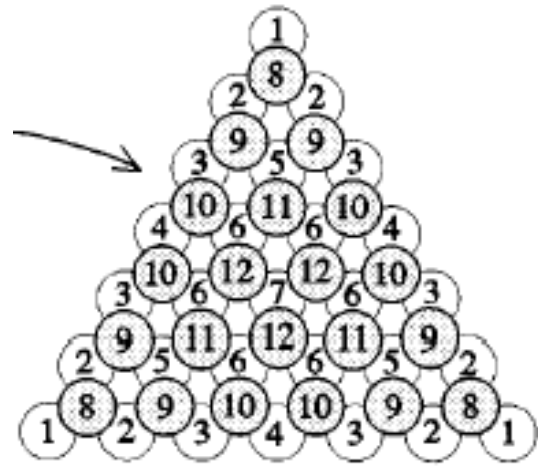


Size dependence

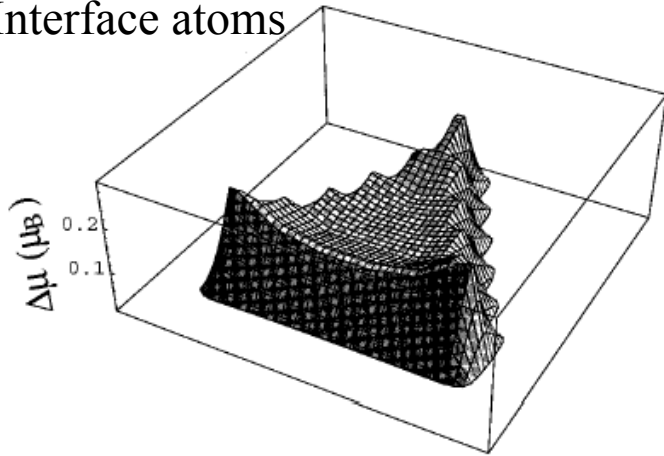
Reduced asymmetry by increasing the cluster size → reduced magnetic moment

Cluster magnetism

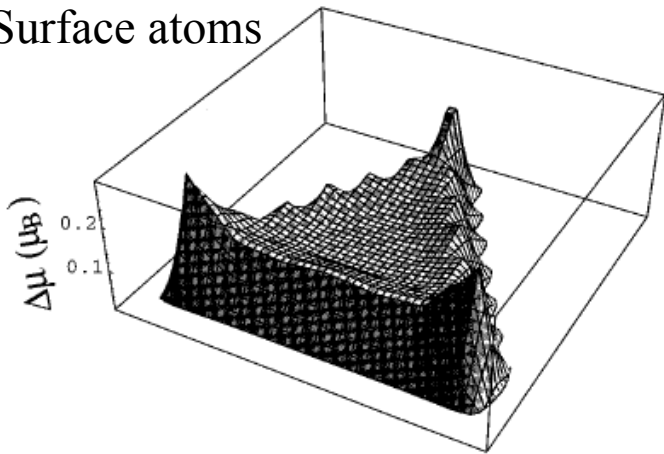
Two atomic layer high Co islands
on Cu(111)



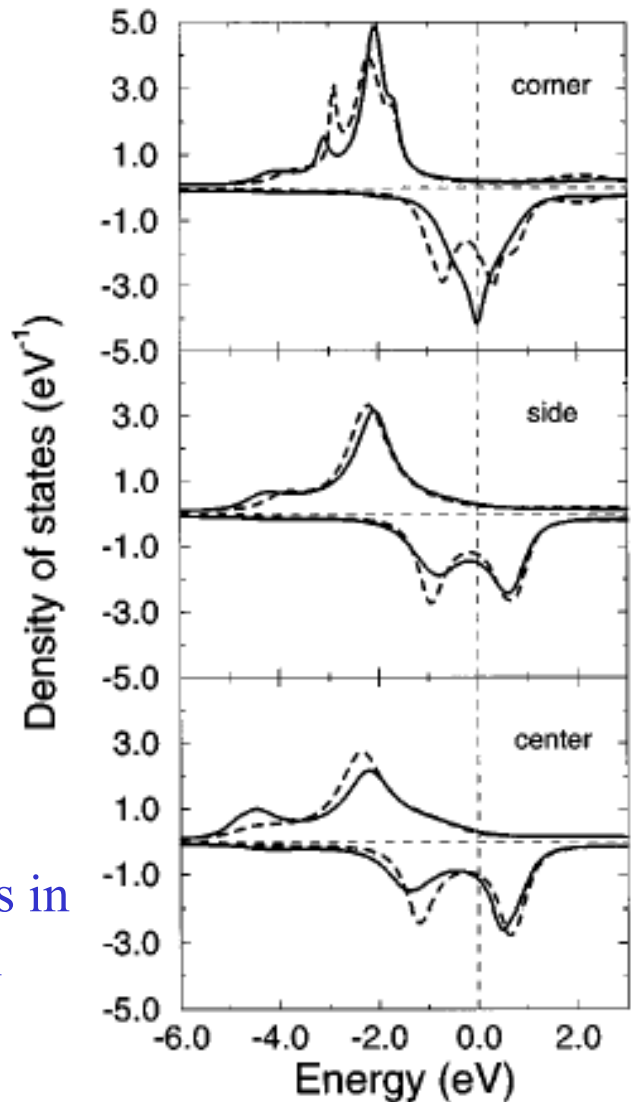
Interface atoms



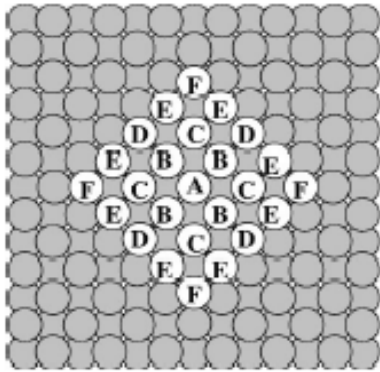
Surface atoms



The magnetic moment of the atoms in
the islands depends on the local
coordination



Cluster magnetism

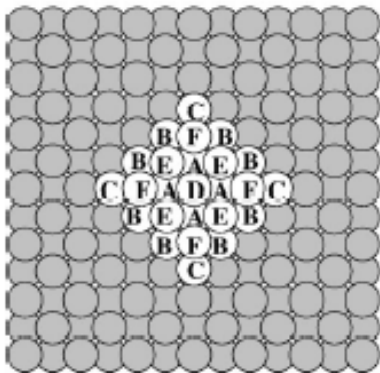


Site	A	B	C	D	E	F
$\mu(\mu_B)$	1.53	1.61	1.66	1.66	1.72	1.80

$$\langle \mu \rangle = 1.69 \mu_B$$

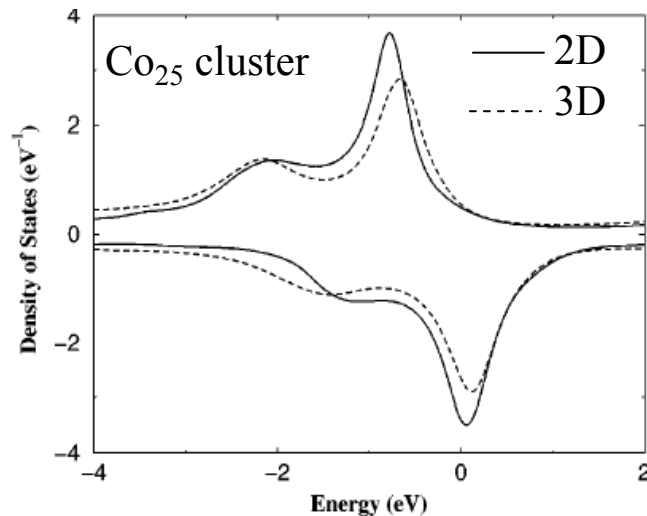
Co/Cu(100)

J. Izquierdo *et al.* Phys. Rev. B
53, 140413 (2001)



Site	A	B	C	D	E	F
$\mu(\mu_B)$	1.15	1.46	1.67	1.52	1.65	1.64

$$\langle \mu \rangle = 1.51 \mu_B$$



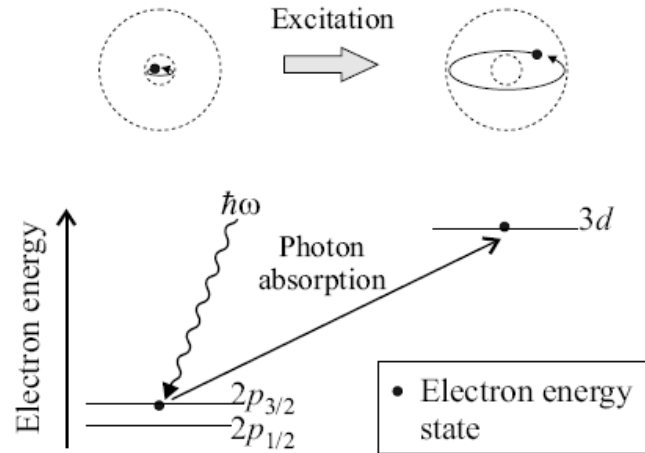
In general:

- Narrowing of the d band at surface \rightarrow increased m_S
- Larger $m_S \rightarrow$ larger m_L due to the spin-orbit coupling
- Symmetry breaking at surface \rightarrow removal of the m_L quenching

XMCD: X-rays Magnetic Circular Dichroism

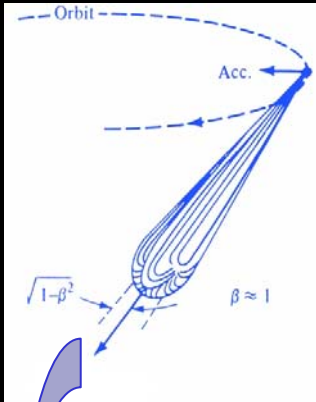
Two step model:

- 1) spin-polarized photoelectrons are created by using circularly polarized x-rays
- 2) these polarized photoelectrons are used to analyze the spin-split final density of states, thus the valence band acts as a spin-sensitive detector.



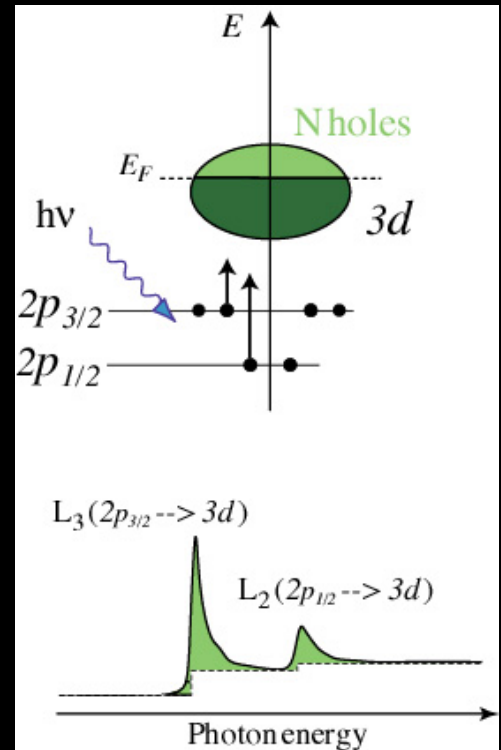
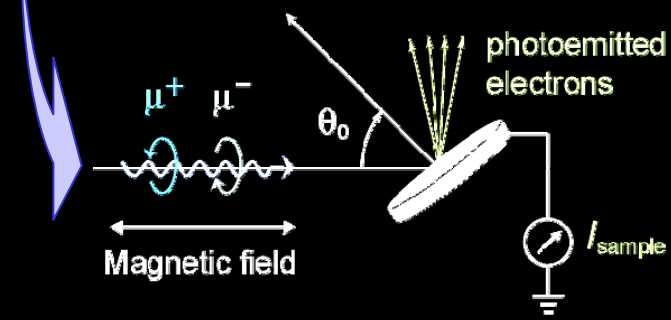
XAS of 3d elements

relativistic electrons



Variable energy
polarized
x-rays

x-ray monochromator



Photons transfer their angular momentum to the excited photoelectrons through the spin-orbit interaction

XMCD theory: basic idea

Photon-electron interaction: dipole approximation

$$H_{\text{int}}(\mathbf{0}, t) \approx \underline{\mathbf{r}} \cdot \underline{\boldsymbol{\varepsilon}}_q [a_k \exp(-i\omega_k t) + \text{c.c.}]$$

The dipole operator $P_q^{(1)} = \underline{\mathbf{r}} \cdot \underline{\boldsymbol{\varepsilon}}_q$ can be written in terms of Racah's tensor operators (where $Y_{l,m}$ are the spherical harmonics)

$$\begin{aligned} q = 0 &\rightarrow \text{linear polarization} & P_{\pm 1}^{(1)} &= \mp \frac{1}{\sqrt{2}} (x \pm iy) = r C_{\pm 1}^{(1)} = r \sqrt{\frac{4\pi}{3}} Y_{1, \pm 1} \\ q = \pm 1 &\rightarrow \text{circular polarization} & P_0^{(1)} &= z = r C_0^{(1)} = r \sqrt{\frac{4\pi}{3}} Y_{1, 0} \end{aligned}$$

The photon absorption generates a transition from an initial core level to a final level close to the Fermi level (note that the spin is not affected)

$$|i\rangle = R_i(r) |c, m_c; s = \frac{1}{2}, m_s\rangle \quad \longrightarrow \quad |f\rangle = R_f(r) |l, m_l; s = \frac{1}{2}, m_s\rangle$$

$\begin{aligned} \langle l, m C_{\pm 1}^{(1)} l-1, m \mp 1 \rangle &= \sqrt{\frac{(l \pm m)(l \pm m - 1)}{2(2l-1)(2l+1)}} \\ &= \sqrt{\frac{l(l-1) \pm (2l-1)m + m^2}{2(2l-1)(2l+1)}} \end{aligned}$	$\begin{aligned} \langle l, m C_{\pm 1}^{(1)} l+1, m \mp 1 \rangle &= -\sqrt{\frac{(l \mp m + 2)(l \mp m + 1)}{2(2l+3)(2l+1)}} \\ &= -\sqrt{\frac{l(l+3) \mp (2l+1)m + m^2}{2(2l+3)(2l+1)}} \end{aligned}$
$\langle l, m C_0^{(1)} l-1, m \rangle = \sqrt{\frac{l^2 - m^2}{(2l-1)(2l+1)}}$	$\langle l, m C_0^{(1)} l+1, m \rangle = \sqrt{\frac{(l+1)^2 - m^2}{(2l+3)(2l+1)}}$

Table 1.3: Electric dipole matrix elements $\langle l, m_l | C_q^{(1)} | c, m_c \rangle$ in the one electron model. The matrix elements are non-vanishing when $c = l - 1$ (left column) or $c = l + 1$ (right column), and when $m_c + q = m_l$. q denotes the state of polarization of the photons which mix the states $|l, m_l\rangle$ and $|c, m_c\rangle$.

Remember: things become easily complicate. For example, if spin-orbit interaction is not negligible the (L, m_L, S, m_S) is not the good basis and you have to use the (L, S, J, m_J) basis

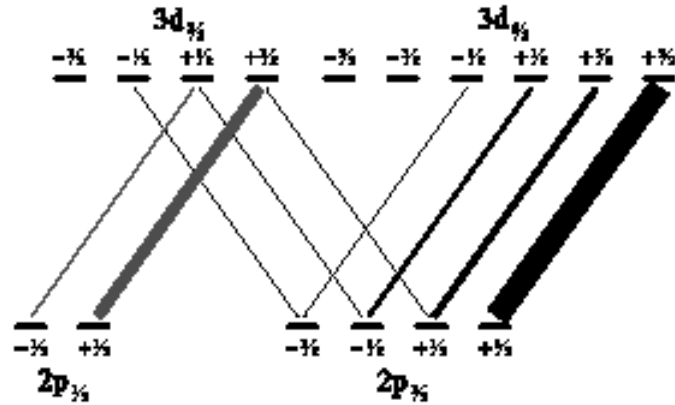
Right ($\Delta m = +1$) and Left ($\Delta m = -1$) circularly polarized photons create photoelectrons with opposite spins.

Since:

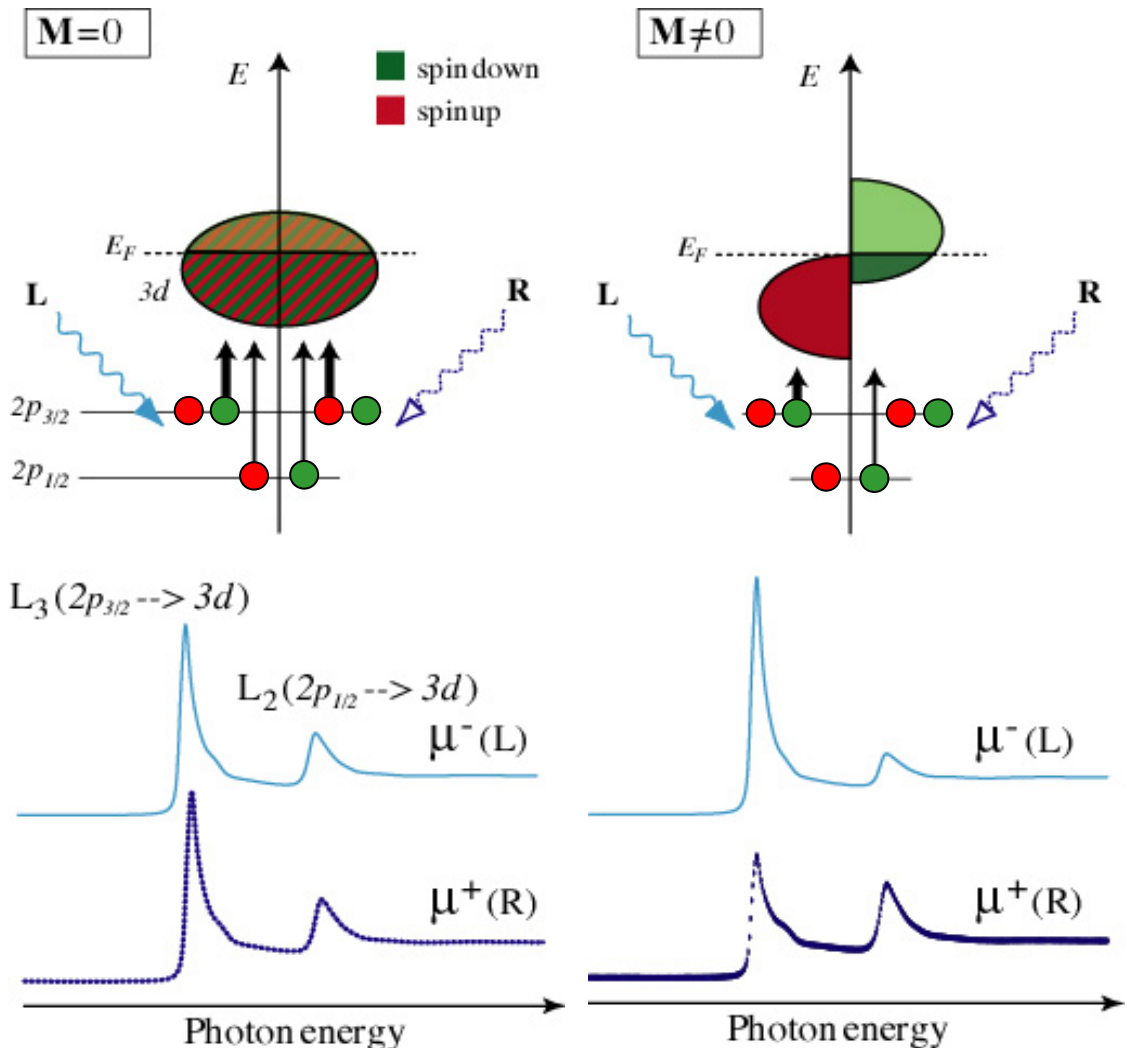
$p_{3/2}$ (L+S coupling)

$p_{1/2}$ (L-S coupling)

The spin polarization of the photoelectrons generated by circularly polarized light is opposite at the L_3 and L_2 edges



Transition probability represented by the line thickness for right circularly polarized x-rays. For clarity the transitions to the 4s levels are neglected (account for less than 5%)



Sum Rules

Separate determination of orbital and spin moment

(actually their projections along the light propagation direction)

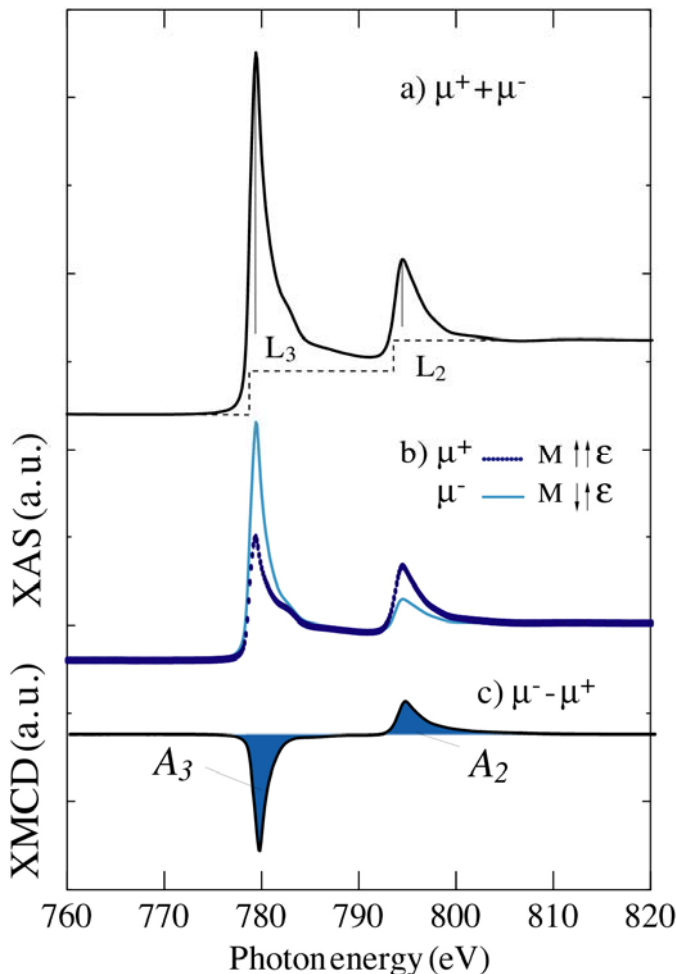
$$m_S = -3 n_h \mu_B (A_3 - 2A_2) + m_T$$

$$m_L = -2 n_h \mu_B (A_3 + A_2)$$

Where A_3 (A_2) is the XMCD measured at the L_3 (L_2) edge, n_h is the number of unoccupied d state (number of holes) and m_T is the magnetic spin dipole moment ($\underline{T} = \underline{S} - 3 \underline{r} (\underline{r} \cdot \underline{S})$).

These rules hold only for itinerant magnetism

30 ML Co/Pt(997)



XMCD: the difference between the XAS spectrum acquired with R and L polarized light

Operatively:

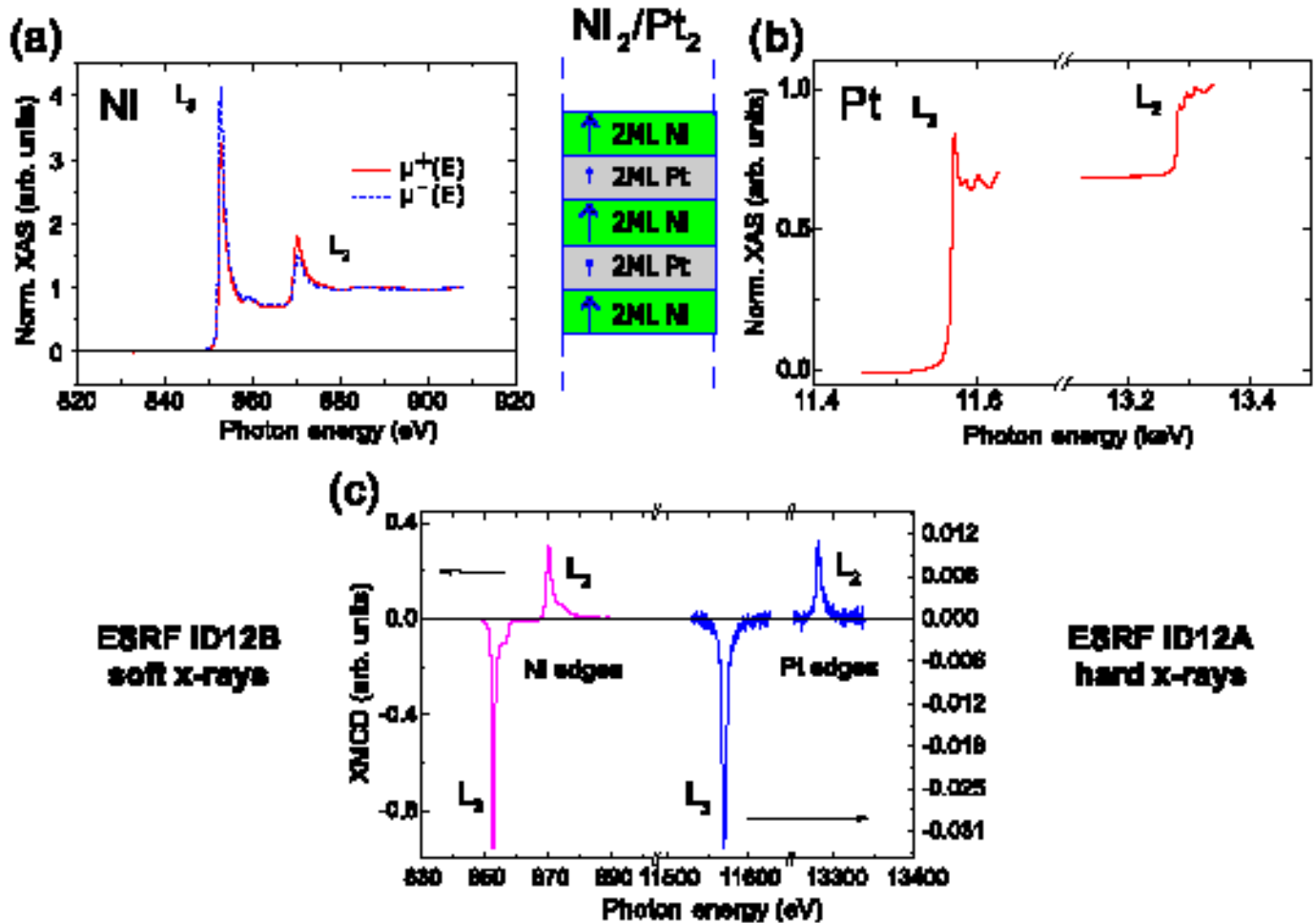
- 1) Measure of the XAS spectra with light polarized R and L in a saturating external magnetic field
- 2) For fcc and bcc crystallographic structures a magic angle ($\sim 57^\circ$) for the photon incidence exists at which $m_T = 0$
- 3) XMCD acquired at different angles gives m_s , m_L , and m_T

B. Thole et al., Phys. Rev. Lett. 68, 1943 (1992)

P. Carra et al., Phys. Rev. Lett. 70, 694 (1993)

J. Stöhr and H. König, Phys. Rev. Lett. 75, 3748 (1995)

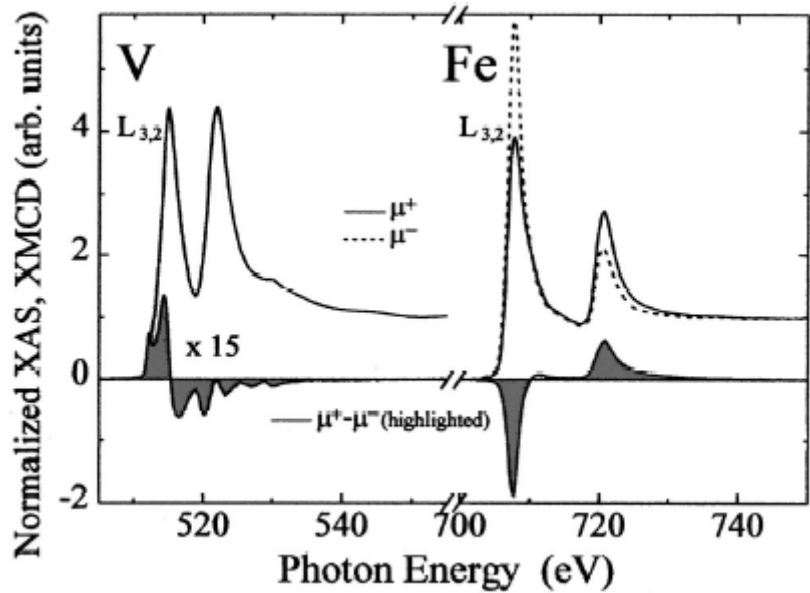
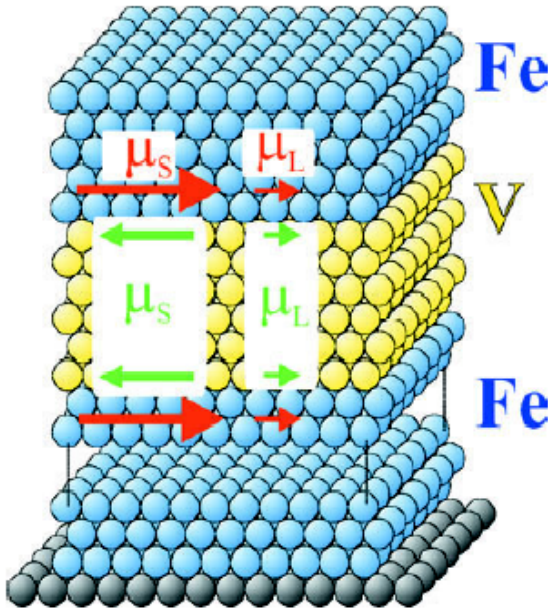
Ni/Pt multilayers



X-ray absorption spectra for a $\text{Ni}_2\text{-Pt}_2$ multilayer: (a) x-ray absorption coefficients for R ($\mu^+(E)$) and L ($\mu^-(E)$) circularly polarized x-rays at the Ni $L_{2,3}$ edges in the soft x-ray range, (b) isotropic XAS at the Pt $L_{2,3}$ edges in the hard x-ray regime, (c) XMCD spectra at the Ni and Pt $L_{2,3}$ edges

XMCD $\neq 0 \rightarrow$ Ni is magnetic (as usual), but also Pt is magnetic

Fe/V/Fe(110) trilayer

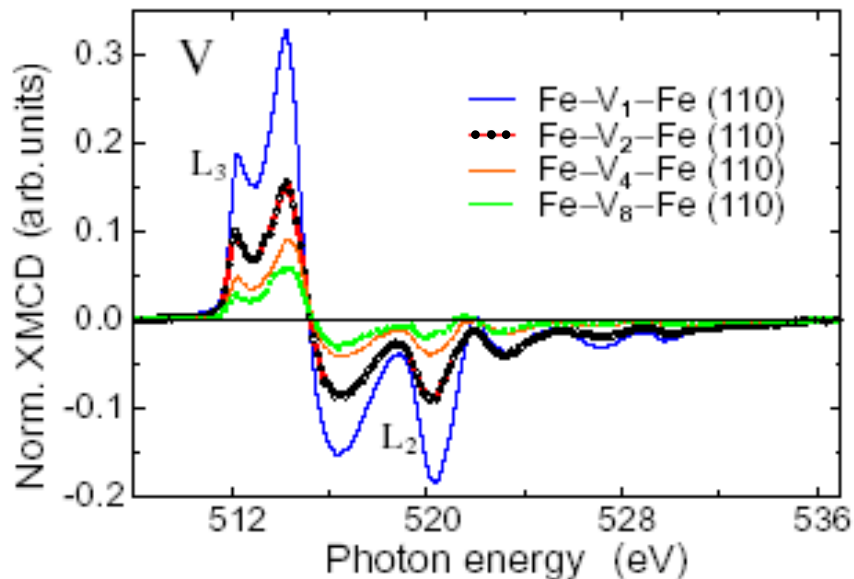


Normalized XAS of L (dashed line) and R (solid line) circularly polarized light and the XMCD at the $L_{2,3}$ edges of V and Fe for a Fe/V4 /Fe(110) trilayer.

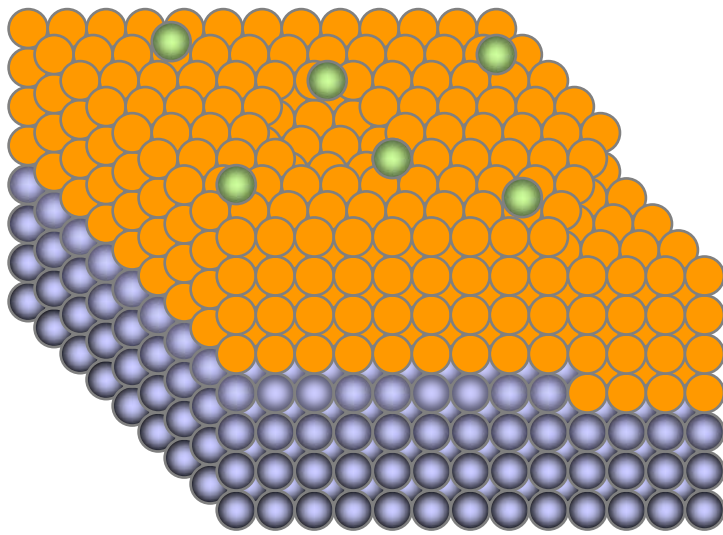
Antiferromagnetic coupling: XMCD signal for V and Fe have opposite signs

The induced magnetic moment in the V atoms strongly reduces with increasing the V thickness

Sum rules can not be applied due to overlapping between $2p^{1/2}$ and $2p^{3/2}$ states



Single atoms on metals surfaces



Fe, Co atoms (<0.01 ML)

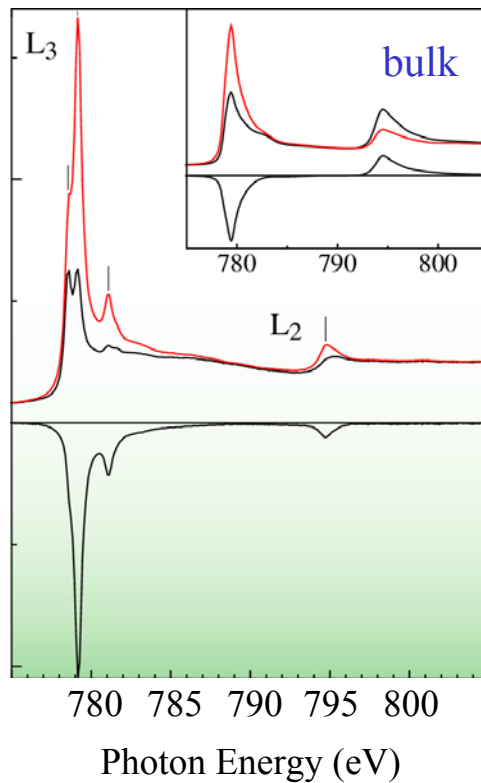
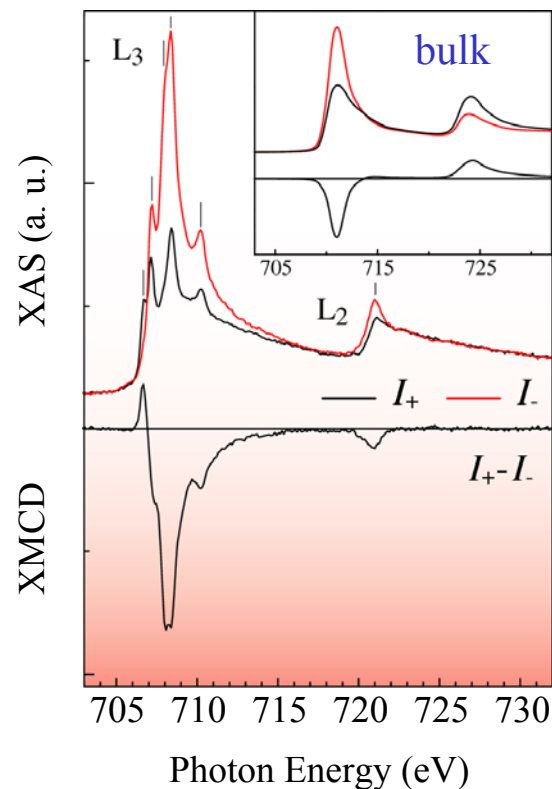
Cs, K, Na, or Li film

Cu(100)

$T = 10 \text{ K}$, $B = 7 \text{ Tesla}$

Fe/K

Co/K



The spectra do not depend on the magnetic field direction \rightarrow

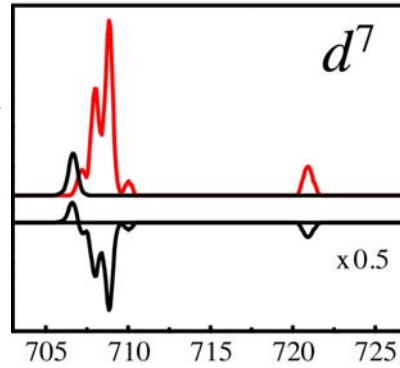
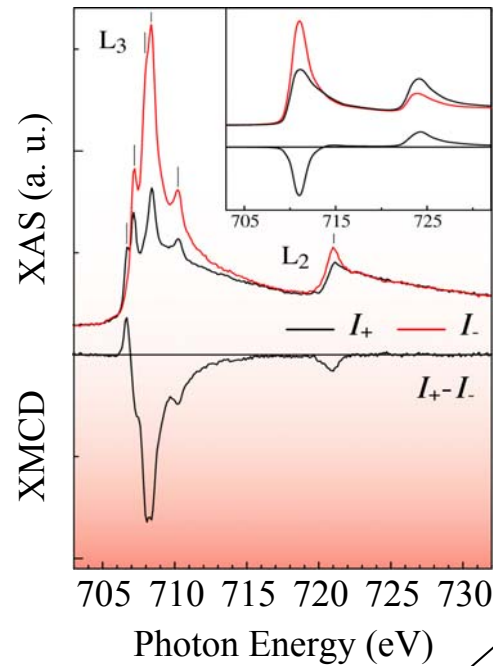
$\mathbf{K} = 0$

Atomic like behavior: soft interaction with the substrate

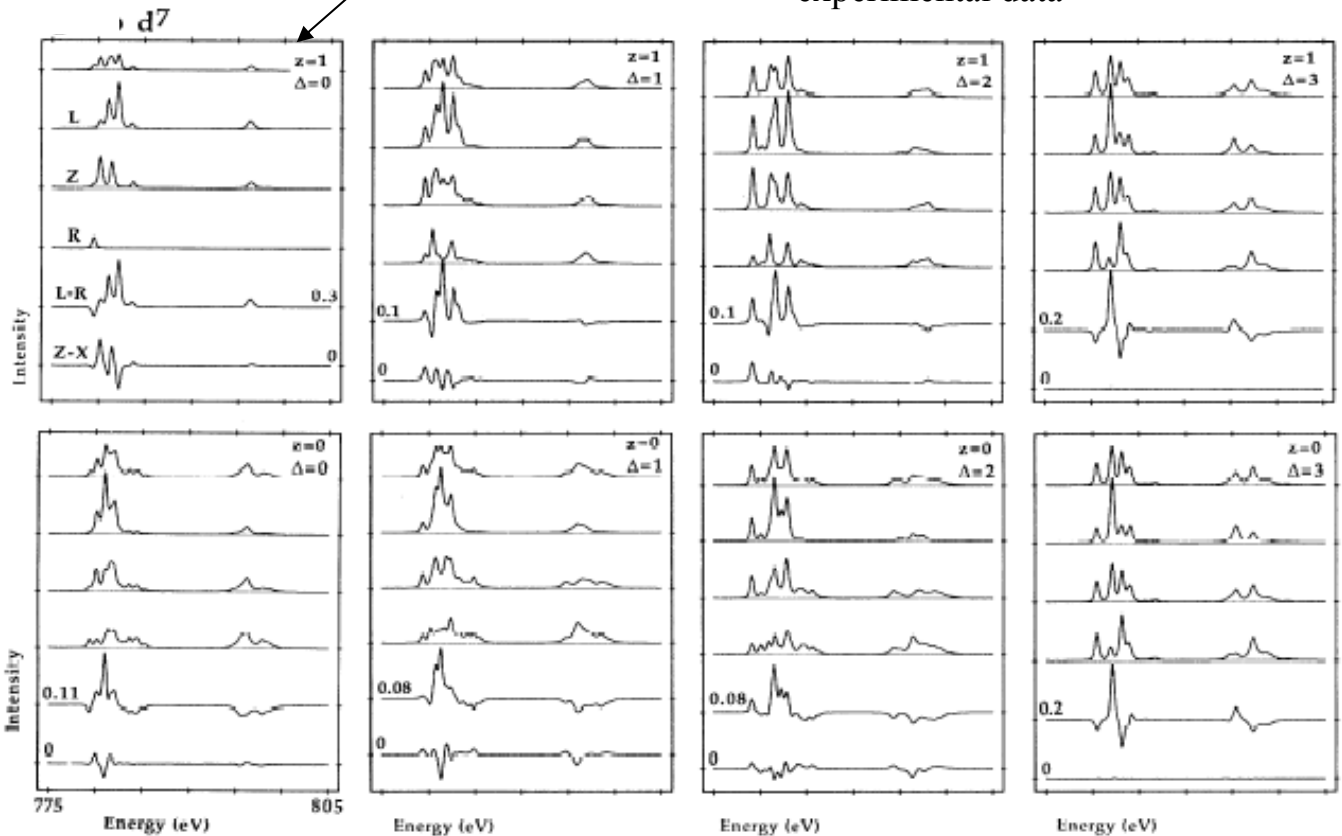
Iron

Fe atom: $d^6 \rightarrow d^7$
 $m_L = 2 \mu_B \rightarrow 3 \mu_B$
 $m_S = 4 \mu_B \rightarrow 3 \mu_B$

Co atom: $d^7 \rightarrow d^8$
 $m_L = 3 \mu_B \rightarrow 3 \mu_B$
 $m_S = 3 \mu_B \rightarrow 2 \mu_B$

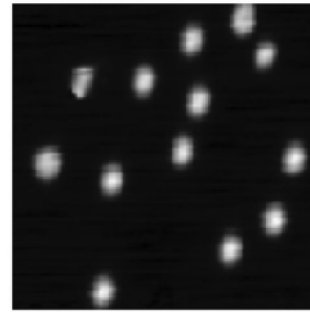
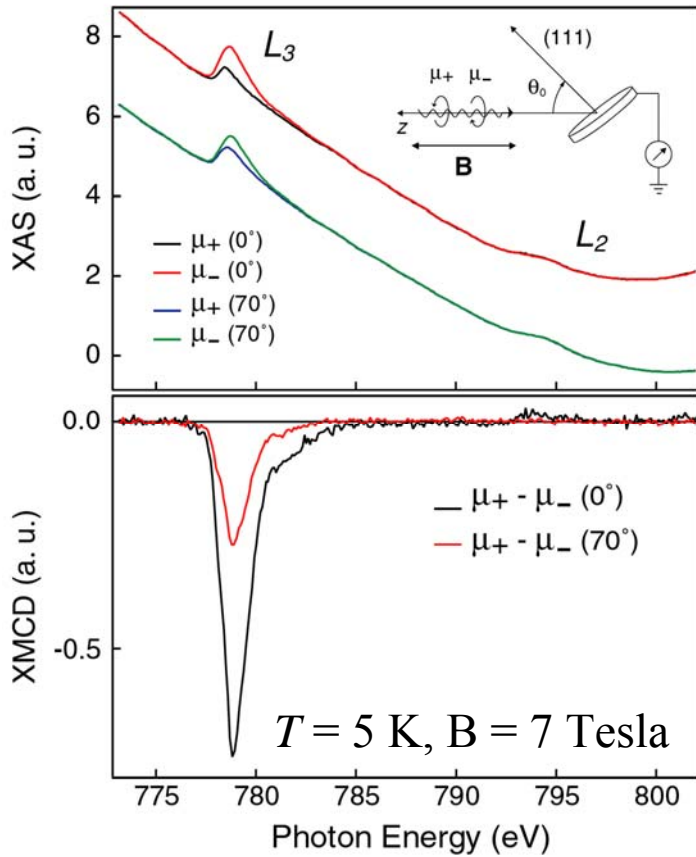


Note: in the simulation L – R is plotted in place of the R – L of the experimental data



XAS spectra for $3d^6 \rightarrow 2p^5 3d^7$ transitions with ($z = 1$) and without ($z = 0$) spin-orbit interactions

Isolated atoms



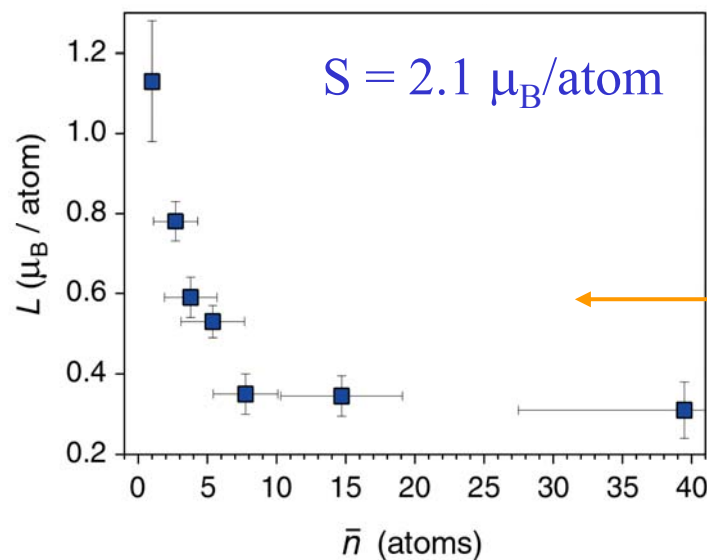
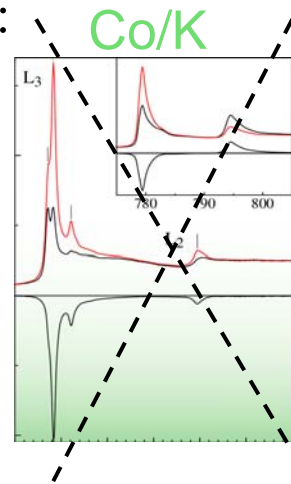
STM image $85 \times 85 \text{ \AA}^2$

The spectrum depends on the magnetic field direction \rightarrow
 $K \neq 0$ (anisotropic system)

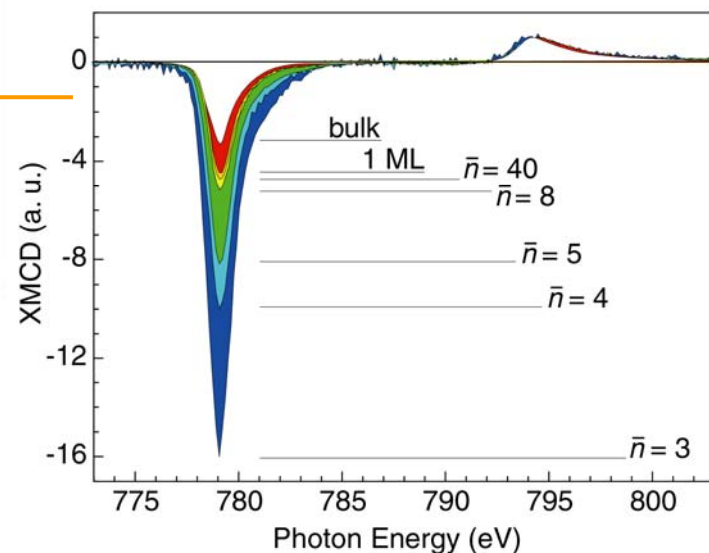
Bulk site spectrum:

sum rules

$$m_L \sim \mu_B (A_3 + A_2)$$



The orbital moment strongly decreases by increasing the cluster size \rightarrow **coordination effect**



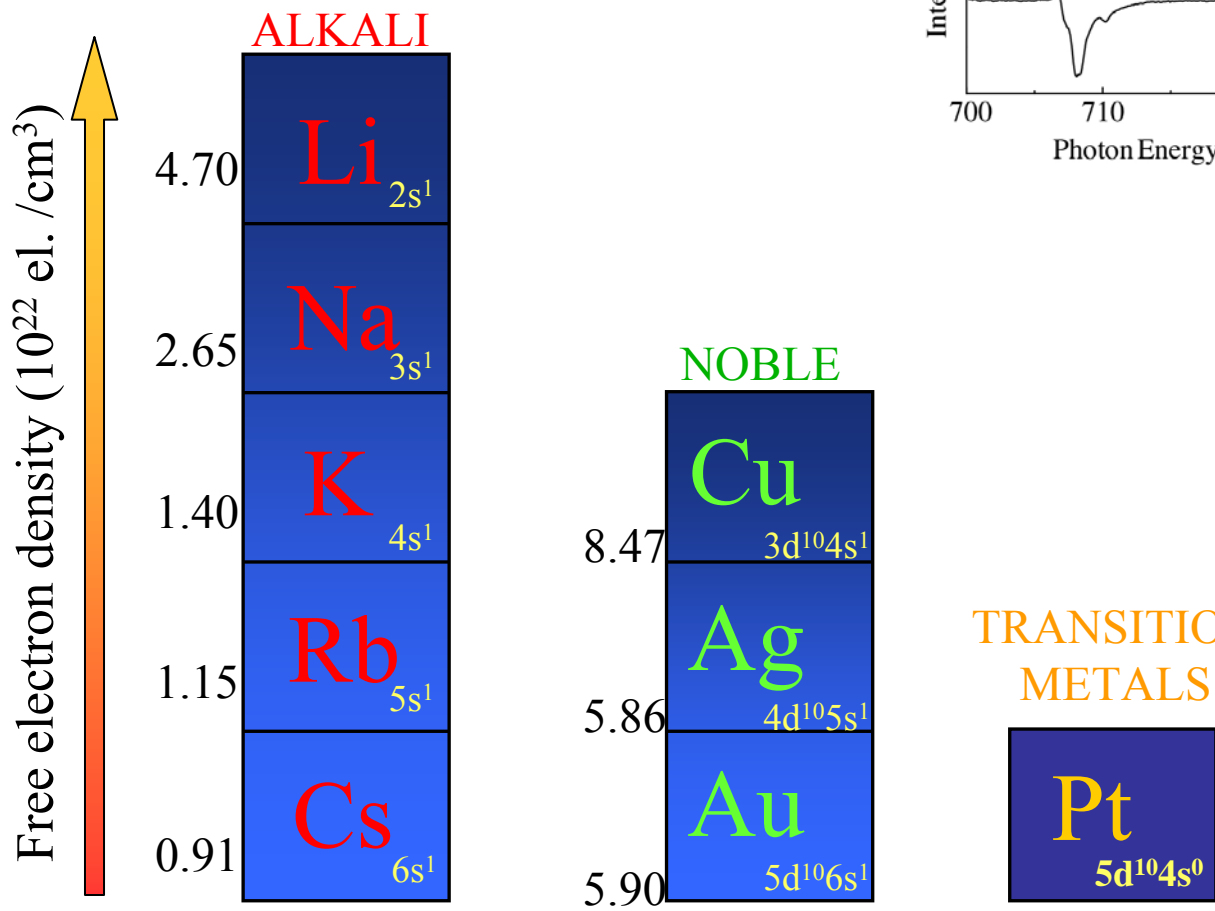
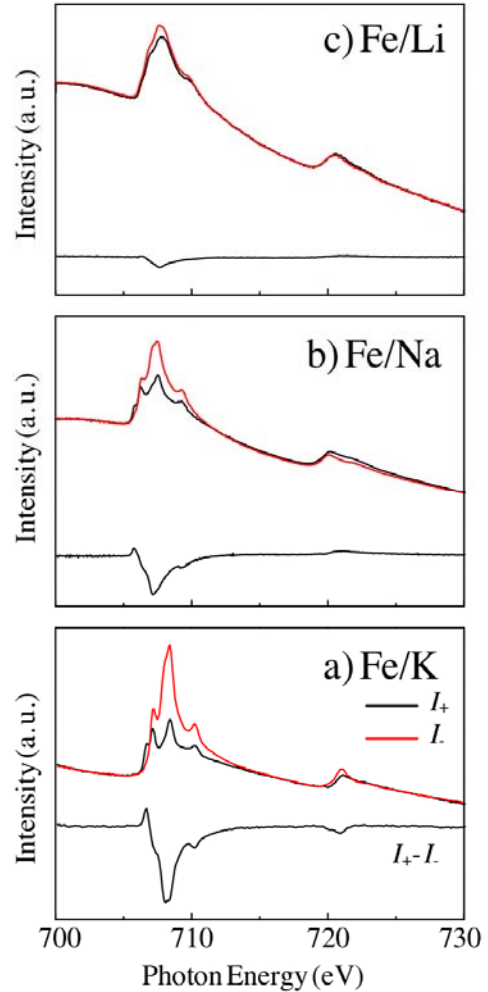
Interaction with the substrate

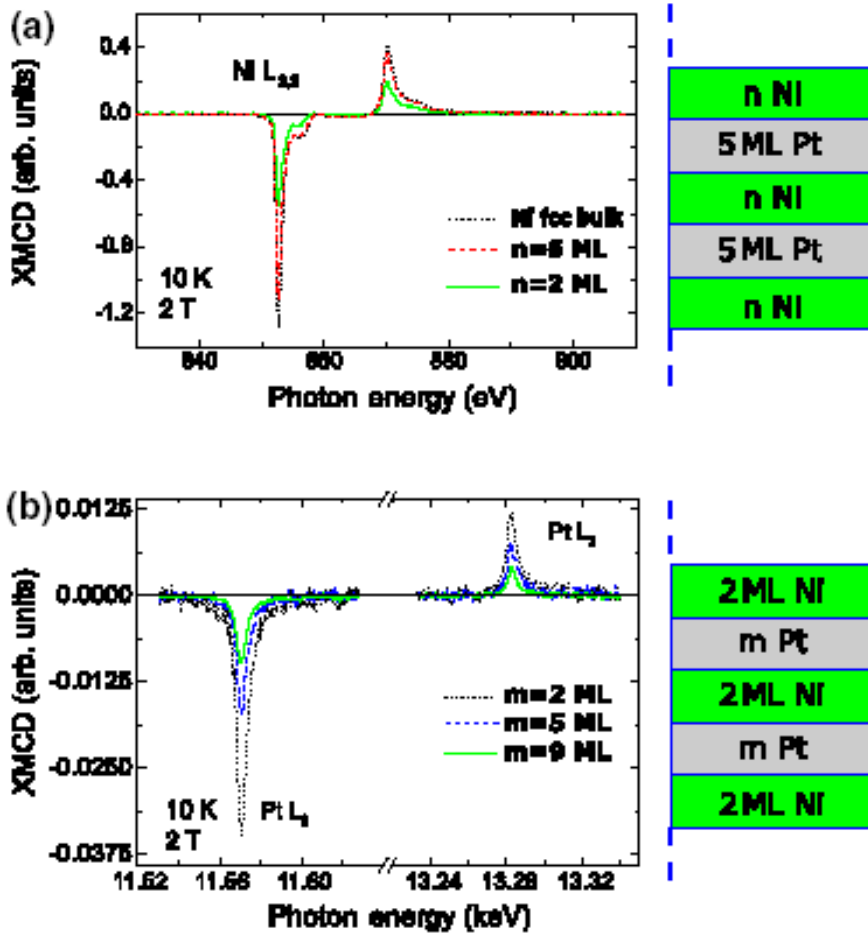
The interaction strength depends on the electronic state of the substrate

The interaction influences both:

- 1) atomic magnetic moment
- 2) Magnetic anisotropy energy

Evolution toward the bulk spectrum





(a) XMCD at the Ni $L_{2,3}$ edges for two multilayers with constant Pt thickness (5 ML) and variable Ni thickness ($\text{Ni}_n\text{-Pt}_5$). (b) Pt $L_{2,3}$ edge XMCD for three multilayers with constant Ni thickness (2 ML) and variable Pt thickness ($\text{Ni}_2\text{-Pt}_m$)

n (ML)	m (ML)	μ_{Ni} ($\mu_{\text{B}}/\text{atom}$)	μ_{Pt} ($\mu_{\text{B}}/\text{atom}$)
2	5	0.24	0.09
2	2	0.39	0.17
6	5	0.47	0.17
6	2	0.49	0.29
13	5	0.54	0.21
Bulk Ni		0.61 [18]	

XMCD data for the average total magnetic moment of Ni and Pt. The induced moments are larger for thin Pt and thick Ni layers. The Ni magnetic moment is always reduced compared to the bulk (see the last row). The effect is stronger for thin Ni separated by thick Pt layers. The error bars are typically of about 10%.