

Wai Kwan LIU, Loris LOMBARDO, Jessica RUFFINER, Andreas ZÜTTEL

Abstract

The gravimetric and volumetric hydrogen density in porous and non-porous nanomaterials is determined with a model only based on two material parameters, the material density and the specific surface area. Materials as small particles after synthesis exhibit a much lower packing density than the crystal density, which significantly reduces the volumetric hydrogen density.

Introduction

Nanomaterials absorb empirically 0.6 monolayer [1] of liq. hydrogen at 100 K based on the surface determined by N₂ adsorption in liq. N₂ with the BET method. However, the fit of the excess hydrogen uptake vs. the BET specific surface area intercepts the axis close to 1.5 mass% H₂ ($A_{\text{spec}} = 0 \text{ m}^2 \cdot \text{g}^{-1}$). The packing density of MOF's is only close to 50% of the crystal density [2].

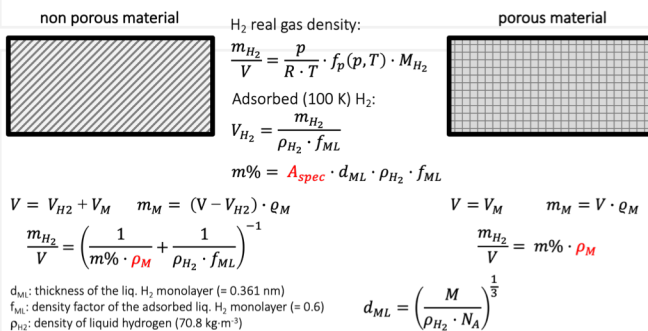


Fig. 1 (above) Model conditions and results for non porous and porous materials. Fig. 2 (right) Volumetric versus gravimetric adsorbed hydrogen density in nanomaterials. The red area indicates the range covered by compressed hydrogen at 298 K.

References

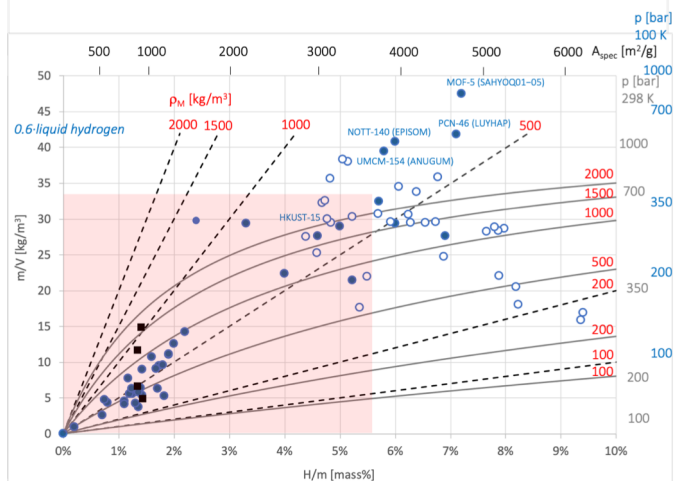
- [1] D.P. Broom, C.J. Webb, G.S. Fanourgakis, G.E. Froudakis, P.N. Trikalitis, M. Hirscher, "Concepts for improving hydrogen storage in nanoporous materials", International Journal of Hydrogen Energy 44 (2019), pp. 7768 – 7779.
- [2] R. Balderas-Xicohtencatl, M. Schlichtenmayer, M. Hirscher, "Volumetric hydrogen storage capacity in metaleorganic frameworks", Energy Technol. 6 (3) (2018), pp. 578-582.
- [3] A. Züttel, W.K. Liu, J. Ruffiner, L. Lombardo, "A heuristic model for H₂ physisorption on and in nanomaterials", to be submitted (2025).
- [4] Poster presentation, GRC: Hydrogen-Metal Systems, Les Diablerets, 22. – 27. June 2025

Conclusion

Very few metal organic frameworks (MOFs) exceed the gravimetric and volumetric hydrogen density at 100 K of compressed H₂ at 700 bar and 298K (red area in fig. 2). However, some reported data are erroneous, the pore volume in the material is not large enough to host the reported amount of hydrogen, due to overestimation of the packing density and the excess hydrogen uptake.

Model

The model assumes the adsorption of 0.6 monolayer of liq. hydrogen at 100 K on the surface of non-porous materials and in the pores of porous materials. The gravimetric and volumetric hydrogen density is calculated with only two material parameter the packing material density ρ_M and the specific surface area A_{spec} using the density of liquid hydrogen ($70.8 \text{ kg} \cdot \text{m}^{-3}$) at 21 K.



Andreas ZÜTTEL, Prof. Dr.

Laboratory of Materials for Renewable Energy (LMER)
Institute of Chemical Sciences and Engineering (ISIC)
Basic Science Faculty (SB)
École polytechnique fédérale de Lausanne (EPFL) Valais/Wallis



Rue de l'Industrie 17, CP 440
CH-1951 Sion, Switzerland

e: andreas.zuetzel@epfl.ch
m: +41 79 484 2553
T: +41 21 695 8304 (Secretary)
U: <http://lmer.epfl.ch>

