7: Two compartment modeling

- 1. What is compartmental modeling?
- 2. How can tracer kinetics be mathematically described?
- 3. How do 2-deoxyglucose methods trace glucose metabolism?

After this course you

- 1. Understand how mass conservation can be used to model tracer kinetics and estimate metabolic rates
- 2. Understand the mathematical principle underlying metabolic modeling of imaging data
- 3. Can apply the principle of modeling tracer uptake to simple kinetic situations
- 4. Understand the basics of modeling deoxyglucose uptake into tissue to extract metabolic rates

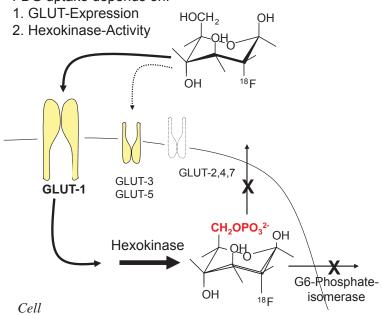
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7-1

How is intracellular glucose metabolism measured?

[¹⁸F]FDG (2-[¹⁸F]Fluoro-2-Deoxy-Glucose)

FDG uptake depends on:



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7-3

7-1. What is a compartment model?

tracers

Definition: Compartment

Concept: Physiological system -

decomposed into N interacting subsystems

Subsystem = chemical species in a physical place (**compartment**)

NB. Tracer is considered to be distributed uniformly in compartment

Inaccessible portion | Recognition | Recogn

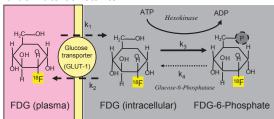
Key elements of compartmental modeling

- 1. Predict inaccessible features of system
- 2. Measurement in the accessible portion
- Estimation of specific parameters of interest.

Steady-state assumption:

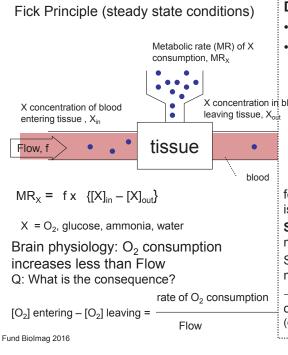
- metabolic rate of process is not changing with time
- 2. concentrations are constant during the evaluation period.

processes can be described with pseudo-firstorder rate constants.



How does conservation of mass allow rate determination?

Fick's principle



Definition Tracer

- · radio-activity emitting, labelled molecule
- structurally related to the natural substance (tracee) or involved in the dynamic process

x concentration in blood See earlier examples, but also O₂ (left)

introduced in a trace amount (=orders of magnitude below tracee); process being measured is not perturbed by it.

few tracer molecules contain radioactive isotope; others contain "cold" isotope

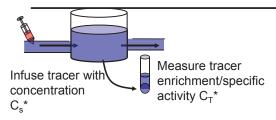
Specific activity (SA) = "hot" / "cold" tracer molecules

SA is always measured; [MBq/µmol or mCi/µmol]

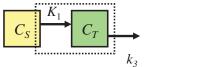
→ convert measured radioactivity concentrations in tissue and blood to mass (correct for physical decay)

7-2. What are first-order tracer kinetics?

One-tissue compartment model



Unidirectional chemical reaction $S \rightarrow T$:



First-order process S->T

Reaction velocity V [µmol/g/min]:

$$V = \frac{dC_T(t)}{dt} = K_1 C_S(t) - k_3 C_T(t)$$

 K_1 , k_3 - (pseudo) first-order rate **constants**;

⇒ independent of concentration and time; unit: [sec-1 or min-1]

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The rate of labeled molecules entering C_T $dC_T^*/dt = Metabolic flux V x probability of$ precursor C_S labeled

$$\frac{dC_T^*(t)}{dt} = V \frac{C_S^*}{C_S} = K_1 C_S^*(t)$$

How many labeled (red) molecules/per min? (Assume the rate is V=10/min)

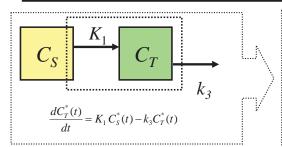
$$\underbrace{\begin{array}{cccc} \circ \circ \circ & \circ \circ \circ \\ \end{array}} \underbrace{\begin{array}{cccc} dC_T^*(t) \\ dt \end{array}} = ?$$

Need to add efflux from C_⊤:

k₃: Metabolic efflux V x probability of molecule C_⊤ being labeled

$$\frac{dC_T^*(t)}{dt} = K_1 C_S^*(t) - k_3 C_T^*(t)$$
7-6

What describes the one-tissue compartment model?

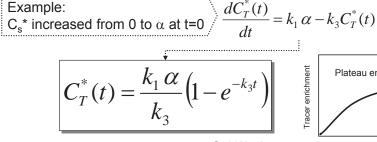


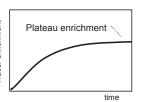
Linear first-order ordinary differential equations (ODEs):

→ Laplace transformation

$$C *_{T} (t) = K_{1}C *_{S} (t) \otimes e^{-k_{3}t}$$

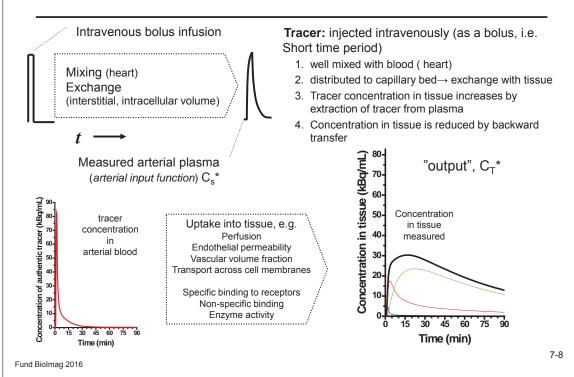
$$a(t) \otimes b(t) = \int_{0}^{t} a(\tau)b(t-\tau)d\tau \dots$$



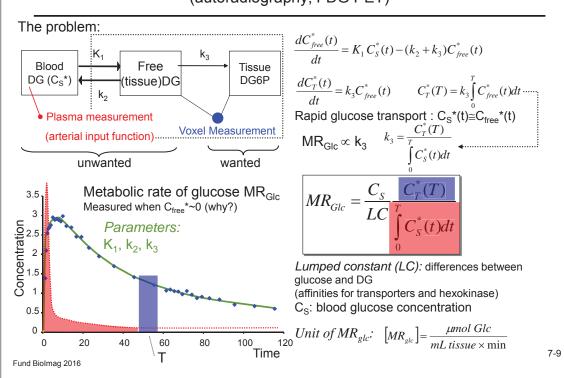


Example:

What is the input function?



7-3. How does Deoxyglucose (DG) measure glucose metabolism ? (autoradiography, FDG PET)



Ex. Typical FDG PET scan

