

## 4 Standard Abbreviations and Acronyms

$\alpha$	observed optical rotation in degrees	compd	compound
$[\alpha]$	specific rotation [expressed without units; the units, (deg·mL)/(g·dm), are understood]	concd	concentrated
Å	angstrom(s)	concn	concentration
Ac	acetyl	COSY	correlation spectroscopy
acac	acetylacetonate	cot	1,3,5,7-cyclooctatetraene
ADP	adenosine 5'-diphosphate	Cp	cyclopentadienyl
AIBN	2,2'-azobisisobutyronitrile	<i>m</i> -CPBA	<i>meta</i> -chloroperoxybenzoic acid
AM1	Austin model 1	CV	cyclic voltammetry
AMP	adenosine 5'-monophosphate	Cy	cyclohexyl
Anal.	combustion elemental analysis	$\delta$	chemical shift in parts per million downfield from tetramethylsilane
anhyd	anhydrous	d	day(s); doublet (spectral); deci
AO	atomic orbital	<i>d</i>	density
aq	aqueous	DABCO	1,4-diazabicyclo[2.2.2]octane
Ar	aryl	dansyl	5-(dimethylamino)- 1-naphthalenesulfonyl
atm	atmosphere(s)	DBN	1,5-diazabicyclo[4.3.0]non-5-ene
ATP	adenosine 5'-triphosphate	DBU	1,8-diazabicyclo[5.4.0]undec-7-ene
ATPase	adenosinetriphosphatase	DCC	<i>N,N'</i> -dicyclohexylcarbodiimide
av	average	DCE	1,2-dichloroethane
9-BBN	9-borabicyclo[3.3.1]nonyl	DDQ	2,3-dichloro-5,6-dicyano- 1,4-benzoquinone
9-BBN-H	9-borabicyclo[3.3.1]nonane	DEAD	diethyl azodicarboxylate
Bn, Bzl	benzyl	DEPT	distortionless enhancement by polarization transfer
bpy	2,2'-bipyridyl	DFT	density functional theory
BOC, Boc	<i>tert</i> -butoxycarbonyl	DIBALH	diisobutylaluminum hydride
bp	boiling point, base pair	DMA	dimethylacetamide
br	broad (spectral)	DMAP	4-( <i>N,N</i> -dimethylamino)pyridine
Bu, <i>n</i> -Bu	normal (primary) butyl	DMDO	dimethyldioxirane
<i>s</i> -Bu	<i>sec</i> -butyl	DME	1,2-dimethoxyethane
<i>t</i> -Bu	<i>tert</i> -butyl	DMF	dimethylformamide
Bz	benzoyl (not benzyl)	DMPU	1,3-dimethyl-3,4,5,6-tetrahydro- 2(1 <i>H</i> )-pyrimidinone
B3LYP	3-parameter hybrid Becke exchange/ Lee–Yang–Parr correlation functional	DMSO	dimethyl sulfoxide
°C	degrees Celsius	DMT	4,4'-dimethoxytrityl (4,4'-dimethoxytriphenylmethyl)
calcd	calculated	DNA	deoxyribonucleic acid
cAMP	adenosine cyclic 3',5'-phosphate	DPS	<i>tert</i> -butyldiphenylsilyl
CAN	ceric ammonium nitrate	dr	diastereomer ratio
CASSCF	complete active space self-consistent field	DTT	dithiothreitol
CASPT2	complete active space with second-order perturbation theory	E1	unimolecular elimination
cat	catalytic	E2	bimolecular elimination
CBZ, Cbz	benzyloxycarbonyl (preferred over the abbreviation Z)	ED <sub>50</sub>	dose effective in 50% of test subjects
CC	coupled cluster	EDTA	ethylenediaminetetraacetic acid
CD	circular dichroism	EI	electron impact
cDNA	complementary deoxyribonucleic acid	EPR	electron paramagnetic resonance
CI	chemical ionization; configuration interaction	eq	equation
CIF	crystallographic information file	equiv	equivalent
CIDNP	chemically induced dynamic nuclear polarization	er	enantiomer ratio
cm	centimeter(s)	ESI	electrospray ionization
cm <sup>-1</sup>	wavenumber(s)	Et	ethyl
cod	1,5-cyclooctadiene	FAB	fast atom bombardment
		FD	field desorption
		FID	flame ionization detector; free induction decay

Fmoc	9-fluorenylmethoxycarbonyl	mM	millimolar (millimoles per liter)
FT	Fourier transform	MO	molecular orbital
g	gram(s); prefix to NMR abbreviation denoting gradient-selected (e.g. gCOSY, gHMQC)	mol	mole(s); molecular (as in mol wt)
GC	gas chromatography	MOM	methoxymethyl
GTP	guanosine 5'-triphosphate	mp	melting point
h	hour(s)	MP	Møller–Plesset perturbation theory
HF	Hartree–Fock	MRCI	multi-reference configuration interaction
HMBC	heteronuclear multiple bond correlation	mRNA	messenger ribonucleic acid
HMPA	hexamethylphosphoric triamide (hexamethylphosphoramide)	Ms	methylsulfonyl (mesyl)
HMQC	heteronuclear multiple quantum correlation	MS	mass spectrometry
HOMO	highest occupied molecular orbital	MTBE	methyl <i>tert</i> -butyl ether
HPLC	high-performance liquid chromatography	MW, mol wt	molecular weight
HRMS	high-resolution mass spectrometry	<i>m/z</i>	mass-to-charge ratio (not <i>m/e</i> )
HSQC	heteronuclear single quantum correlation	N	normal (equivalents per liter)
Hz	hertz	NAD <sup>+</sup>	nicotinamide adenine dinucleotide
ICR	ion cyclotron resonance	NADH	reduced NAD
INDO	intermediate neglect of differential overlap	NBO	natural bond orbital
IP	ionization potential	NBS	<i>N</i> -bromosuccinimide
IR	infrared	NCS	<i>N</i> -chlorosuccinimide
<i>J</i>	coupling constant (in NMR spectrometry)	NICS	nucleus-independent chemical shift
k	kilo	nm	nanometer(s)
K	kelvin(s) (absolute temperature)	NMO	<i>N</i> -methylmorpholine- <i>N</i> -oxide
L	liter(s)	NMP	<i>N</i> -methylpyrrolidone
LAH	lithium aluminum hydride	NMR	nuclear magnetic resonance
LCAO	linear combination of atomic orbitals	NOE	nuclear Overhauser effect
LD <sub>50</sub>	dose that is lethal in 50% of test subjects	NOESY	nuclear Overhauser effect spectroscopy
LDA	lithium diisopropylamide; local density approximation	NRT	natural resonance theory
LFER	linear free energy relationship	Nu	nucleophile
LHMDS	lithium hexamethyldisilazane, lithium bis(trimethylsilyl)amide	obsd	observed
lit.	literature value (abbreviation used with period)	OD	optical density
LTMP	lithium 2,2,6,6-tetramethylpiperidide	ORD	optical rotary dispersion
LUMO	lowest unoccupied molecular orbital	PCC	pyridinium chlorochromate
μ	micro	PDC	pyridinium dichromate
m	multiplet (spectral); meter(s); millimolar (moles per liter); mega	PES	photoelectron spectroscopy
M	parent molecular ion	Ph	phenyl
M <sup>+</sup>	parent molecular ion	piv	pivaloyl
MALDI	matrix-assisted laser desorption ionization	pm	picometer(s)
max	maximum	PM3	parametric method 3
MCD	magnetic circular dichroism	PMB	<i>p</i> -methoxybenzyl
MCR	multicomponent reaction	PPA	poly(phosphoric acid)
MCSCF	multi-configuration self-consistent field	ppm	part(s) per million
MD	molecular dynamics	PPTS	pyridinium <i>para</i> -toluenesulfonate
Me	methyl	Pr	propyl
MEM	(2-methoxyethoxy)methyl	<i>i</i> -Pr	isopropyl
Mes	2,4,6-trimethylphenyl (mesityl) [not methylsulfonyl (mesyl)]	PT	perturbation theory
MHz	megahertz	PTC	phase-transfer catalysis
min	minute(s); minimum	py	pyridine
		q	quartet (spectral)
		QSAR	quantitative structure–activity relationship
		RCM	ring-closure metathesis
		redox	reduction–oxidation
		rel	relative
		<i>R<sub>f</sub></i>	retention factor (in chromatography)
		RHF	restricted Hartree–Fock
		ROESY	rotating frame Overhauser effect spectroscopy
		ROMP	ring-opening metathesis polymerization
		rRNA	ribosomal ribonucleic acid

rt	room temperature	TFA	trifluoroacetic acid
s	singlet (spectral); second(s)	TFAA	trifluoroacetic anhydride
SAR	structure–activity relationship	THF	tetrahydrofuran
SCF	self-consistent field	THP	tetrahydropyran-2-yl
SEM	scanning electron microscopy; 2-trimethylsilylethoxymethyl	TIPS	triisopropylsilyl
SET	single electron transfer	TLC	thin-layer chromatography
S <sub>N</sub> 1	unimolecular nucleophilic substitution	TMAI	tetramethylammonium iodide
S <sub>N</sub> 2	bimolecular nucleophilic substitution	TMEDA	<i>N,N,N',N'</i> -tetramethyl- 1,2-ethylenediamine
S <sub>N</sub> '	nucleophilic substitution with allylic rearrangement	TMS	trimethylsilyl; tetramethylsilane
SOMO	single-occupied molecular orbital	TOF	time-of-flight
t	triplet (spectral)	Tr	triphenylmethyl (trityl)
<i>t</i>	time; temperature in units of degrees Celsius (°C)	tRNA	transfer ribonucleic acid
<i>T</i>	absolute temperature in units of kelvins (K)	<i>t<sub>R</sub></i>	retention time (in chromatography)
TBAB	tetrabutylammonium bromide	Ts	<i>para</i> -toluenesulfonyl (tosyl)
TBAC	tetrabutylammonium chloride	TS	transition state
TBAF	tetrabutylammonium fluoride	UHF	unrestricted Hartree–Fock
TBS	<i>tert</i> -butyldimethylsilyl	UV	ultraviolet
TBHP	<i>tert</i> -butyl hydroperoxide	VCD	vibrational circular dichroism
TCA	trichloroacetic acid	vis	visible
TCNE	tetracyanoethylene	vol	volume
TDDFT	time-dependent density functional theory	v/v	volume per unit volume (volume-to-volume ratio)
TEAB	tetraethylammonium bromide	wt	weight
temp	temperature	w/w	weight per unit weight (weight-to-weight ratio)
Tf	trifluoromethanesulfonyl (triflyl)	ZINDO	Zerner parameterization of intermediate neglect of differential overlap